

Colligative Properties

Colligative properties are characteristics of a solution that depend on the ratio of the number of solute particle to solvent particles.

Freezing
Point
Depression



Boiling
Point
Elevation



Osmotic
Pressure



Vapor
Pressure
Lowering



Solutions And Colligative Properties

DR AJAY



Solutions And Colligative Properties:

Chemistry and Our Universe, Certain properties of solutions depend only on the concentration of the solute particles dissolved not on the nature of the particles Called colligative properties these involve such behaviors as lowering the freezing point raising the boiling point and osmotic pressure Study examples of each **A Textbook of Physical Chemistry** A. S. Negi, S. C. Anand, 1985 Written primarily to meet the requirements of students at the undergraduate level this book aims for a self learning approach The fundamentals of physical chemistry have been explained with illustrations diagrams tables experimental techniques and solved problems **General Chemistry** S. Ekambaram, 2012 General Chemistry presents the fundamental concepts of general chemistry in a precise and comprehensive manner for undergraduate students of chemistry and life science at all Indian universities Adhering strictly to the UGC curriculum the contents are written in a simple and lucid language enriched with a large number of examples and illustrations **Solutions** Wilhelm Ostwald, 1891

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