

# INTERNAL COMBUSTION ENGINE FUNDAMENTALS

SECOND EDITION

## Solution manual



JOHN B. HEYWOOD



# Heywood Solution Manual

**Marcel A. Müller**



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$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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web the net ionic equation for equation 1 is given as equation 3  $\text{s}_2 \text{ aq}$   $\text{fe}_2 \text{ aq}$   $\text{gt}$   $\text{fes s}$  eq 3 8 note that the phase indicator aq is optional on ionic species but is

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web  $\text{cu clo}_4$  2 is soluble all perchlorates are soluble while  $\text{pbso}_4$  is insoluble sulfates are generally soluble except for  $\text{pb}^2$   $\text{ba}^2$   $\text{hg}_2^2$  and  $\text{sr}^2$  the net ionic equation is

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