

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Matching

Match each item with the correct statement below.

Match each item with the correct statement below.

- | | |
|--------------------------------|--------------------|
| a. monatomic ion | f. cation |
| b. acid | g. binary compound |
| c. base | h. anion |
| d. law of definite proportions | i. polyatomic ion |
| e. law of multiple proportions | |

- ___ 1. consists of a single atom with a positive or negative charge
- ___ 2. atom or group of atoms having a negative charge
- ___ 3. atom or group of atoms having a positive charge
- ___ 4. tightly-bound group of atoms that behaves as a unit and carries a net charge
- ___ 5. compound composed of two different elements
- ___ 6. produces a hydrogen ion when dissolved in water
- ___ 7. produces a hydroxide ion when dissolved in water
- ___ 8. In any chemical compound, the masses of elements are always in the same proportion by mass.
- ___ 9. when two elements form more than one compound, the masses of one element that combine with the same mass of the other element are in the ratio of small, whole numbers

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ___ 10. When naming a transition metal ion that can have more than one common ionic charge, the numerical value of the charge is indicated by a ____.
- | | |
|-----------|-------------------------------------|
| a. prefix | c. Roman numeral following the name |
| b. suffix | d. superscript after the name |
- ___ 11. Which of the following correctly provides the name of the element, the symbol for the ion, and the name of the ion?
- | | |
|-----------------------------------|-------------------------------------|
| a. fluorine, F^+ , fluoride ion | c. copper, Cu^+ , cuprous ion |
| b. zinc, Zn^{2+} , zincate ion | d. sulfur, S^{2-} , sulfurous ion |
- ___ 12. The nonmetals in Groups 6A and 7A ____.
- lose electrons when they form ions
 - have a numerical charge that is found by subtracting 8 from the group number
 - all have ions with a -1 charge
 - end in *-ite*
- ___ 13. Which of the following is NOT a cation?
- | | |
|------------------|------------------|
| a. iron(III) ion | c. Ca^{2+} |
| b. sulfate | d. mercurous ion |

Chapter 9 Practice Test Naming And Writing Chemical Formulas

Daniel F McAuley



Chapter 9 Practice Test Naming And Writing Chemical Formulas:

Chapter 9 Practice Test Naming And Writing Chemical Formulas Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Chapter 9 Practice Test Naming And Writing Chemical Formulas**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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