

First/Second Semester B.E. Degree Examination, June/July 2013
Engineering Physics

Time: 3 hrs.

Max. Marks:100

Note: 1. Answer any FIVE full questions, choosing at least two from each part.

2. Answer all objective type questions only on OMR sheet page 5 of the answer booklet.

3. Answer to objective type questions on sheets other than OMR will not be valued.

4. Physical Constants : Planck's constant, $h = 6.63 \times 10^{-34}$ Js, Electron charge, $e = 1.6 \times 10^{-19}$ C
 Electron mass, $m = 9.11 \times 10^{-31}$ kg, Velocity of light, $C = 3 \times 10^8$ mS⁻¹

PART - A

1. a. Choose the correct answers for the following : (04 Marks)
- If red and blue stars emits radiations of continuous wavelengths, then according to Wien's displacement law.
 - Blue star is hotter than red star
 - Red star is hotter than blue star
 - Both stars are at same temperature
 - Difficult to conclude.
 - The expression for de-Broglie wavelength for an electron under an accelerating potential V is,
 - $\frac{12.26}{\sqrt{V}}$ m
 - $\frac{12.26}{\sqrt{V}}$ A°
 - $\frac{12.26}{\sqrt{V}}$ nm
 - $\frac{12.26}{\sqrt{V}}$ μm
 - A particle moves with velocity 3×10^8 ms⁻¹. The wavelength associated with it is 1 nm. Then group velocity of the particle is,
 - 3×10^8 mS⁻¹
 - 3×10^{16} mS⁻¹
 - 3×10^8 mS⁻¹
 - 1.5×10^8 mS⁻¹
 - According to the Compton effect, the wavelength of X-rays scattered at an angle greater than zero,
 - Decreases
 - Doesn't change
 - Increases
 - None of these
- b. Derive an expression for group velocity on the basis of superposition of waves. Also obtain the relation between group velocity and phase velocity. (08 Marks)
- c. Show that Planck's law reduces to Wien's law and Rayleigh-Jeans law under certain conditions. (05 Marks)
- d. Calculate the de-Broglie wavelength associated with an electron of energy 1.5 eV. (03 Marks)
2. a. Choose the correct answers for the following : (04 Marks)
- The energy of the lowest state in one dimensional potential box of length $a = 1$ unit is,
 - $\frac{h^2}{8m}$
 - zero
 - $\frac{h^2}{4ma^2}$
 - $\frac{h^2}{2ma^2}$
 - For a particle which is not bound to any system and is free, the energy eigen value is,
 - zero
 - finite but not quantized
 - infinity
 - finite but quantized
 - If the uncertainty in the position of a particle is equal to its de-Broglie wavelength then uncertainty in its momentum will be,
 - $\Delta P \geq \frac{h}{4\pi}$
 - $\Delta P \geq \frac{h}{2\pi}$
 - $\Delta P \geq \frac{P}{4\pi}$
 - $\Delta P \geq \frac{h}{P}$
 - For an electron to be present inside the nucleus of an atom the uncertainty in the position of the electron must be,
 - more than or equal to the radius of the nucleus
 - more than or equal to the diameter of the nucleus.
 - more than the diameter of the nucleus
 - less than or equal to the diameter of the nucleus.



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