

## Section Review 8.1

### Part A Completion

- |                      |              |
|----------------------|--------------|
| 1. compound          | 6. low       |
| 2. molecular         | 7. high      |
| 3. nonmetals         | 8. atoms     |
| 4. diatomic          | 9. structure |
| 5. molecular formula |              |

### Part B True-False

- |        |        |        |
|--------|--------|--------|
| 10. ST | 12. AT | 14. ST |
| 11. NT | 13. NT |        |

### Part C Matching

- |       |       |       |
|-------|-------|-------|
| 15. e | 17. d | 19. c |
| 16. a | 18. b |       |

### Part D Questions and Problems

20. molecular compound
21. a. 4 carbon atoms, 10 hydrogen atoms  
b. 6 carbon atoms, 5 hydrogen atoms, 1 fluorine atom
22. a. molecule      d. atom  
b. atom            e. molecule  
c. molecule

## Section Review 8.2

### Part A Completion

- stable electron
- covalent
- shared
- single
- unshared pairs
- double/triple
- coordinate covalent bond
- Energy
- bond dissociation energy
- resonance structure

### Part B True-False

- |        |        |        |
|--------|--------|--------|
| 11. NT | 13. AT | 15. ST |
| 12. NT | 14. AT | 16. NT |

### Part C Matching

- |       |       |       |
|-------|-------|-------|
| 17. e | 19. b | 21. d |
| 18. a | 20. c |       |

## Part D Questions and Problems

22. a.  $\text{:}\ddot{\text{Br}}\text{:}\ddot{\text{Br}}\text{:}$
- b.  $\text{H}:\text{C}:::\text{N}:$
- c.  $\begin{array}{c} \text{H} \\ | \\ \text{H}:\text{N}:\text{H} \\ | \\ \text{H} \end{array}$

## Section 8.3

### Part A Completion

- molecular orbitals
- bonding orbital
- lower
- sigma or  $\sigma$
- pi or  $\pi$
- three-dimensional
- VSEPR theory
- orbital hybridization

### Part B True-False

- |        |        |        |
|--------|--------|--------|
| 9. AT  | 11. NT | 13. ST |
| 10. ST | 12. AT | 14. AT |

### Part C Matching

- |       |       |       |
|-------|-------|-------|
| 15. d | 17. a | 19. c |
| 16. e | 18. b |       |

### Part D Questions and Problems

20. Reading from left to right:  
 $sp^3$   $sp^2$   $sp^2$   $sp$   $sp$   $sp^3$

## Section 8.4

### Part A Completion

- equally
- nonpolar
- unequally
- polar
- electronegativities
- dipole interactions
- hydrogen bond
- electronegative
- oxygen, nitrogen, or fluorine

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