

PHYSICAL CHEMISTRY

(Complete Syllabus)

SECTION - A

Straight Objective Type

This section contains 220 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- The mass of oxygen required to produce enough CO which completely reduces x gm of Fe_2O_3 . Suppose molecular weight of $\text{O}_2 = y$, molecular weight of $\text{Fe}_2\text{O}_3 = z$
(1) $\frac{2xz}{3y}$ (2) $\frac{3xy}{2z}$
(3) $\frac{2y}{3xz}$ (4) $\frac{3z}{2xy}$
- The number of lone pair of electrons present in 2.56 gm of orthorhombic sulphur will be
(1) $0.02 N_0$ (2) $0.16 N_0$
(3) $0.32 N_0$ (4) $0.08 N_0$
- A 10.0 g of sample of mixture of calcium chloride and sodium chloride is treated with Na_2CO_3 to precipitate the calcium as calcium carbonate. This CaCO_3 is heated to convert all calcium to CaO and final mass of CaO is 1.62 gms. The % by mass of CaCl_2 in the original mixture is
(1) 15.2%
(2) 32.1%
(3) 21.8%
(4) 11.07%
- 1000 ml of oxygen at STP were passed through an ozoniser and the resulting volume was 888 ml at STP. This quantity of ozonised oxygen is passed through excess of KI solution. The weight of I_2 liberated is
(1) 25.4 g (2) 12.7 g
(3) 2.54 g (4) 1.27 g
- x_1 ml solution having normality 1 is diluted to get a solution of normality y_2 . What would be volume required for dilution?
(1) $x_1 \left(1 - \frac{y_1}{y_2}\right)$ (2) $x_1 \left(\frac{y_2 - 1}{y_1}\right)$
(3) $x_1 \left(\frac{y_1 - 1}{y_2}\right)$ (4) $x_1 \left(1 + \frac{y_1}{y_2}\right)$
- 1.625 g of an interhalogen compound ICl_x was subjected to electric discharge which split it into constituents and brought to STP. It occupied 112 ml. The interhalogen compound is
(1) ICl (2) ICl_3
(3) ICl_5 (4) ICl_7
- If 12.25 g of a sample of KClO_3 of 75% purity and rest vapourisable impurities is strongly heated to give KCl , KClO_4 and 0.32 gm O_2 . The mass of residue obtained is
(1) 1.769 g (2) 7.096 g
(3) 5.327 g (4) 8.85 g
- A 2 g sample of xenon reacts with fluorine. The mass of compound produced is 3.158 g. The empirical formula of the compound is
(1) XeF_2 (2) XeF_4
(3) XeF_6 (4) XeF_8
- 1 g of carbonate of metal was dissolved in 25 ml of 1 N HCl. The resulting liquid required 5 ml of 1 N NaOH for neutralisation. The equivalent weight

Physical Chemistry Test Bank Questions With Answer

A Loxley



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Fundamentals of Physical Chemistry Ananya Ganguly, 2011 Fundamentals of Physical Chemistry is the signature compilation of the class tested notes of iconic chemistry coach Ananya Ganguly Her unique teaching methodology and authoritative approach in teaching of concepts their application and strategy is ideal for preparing for the IITJEE examinations The author s impeccable command and the authority on each foray of chemistry teaching are visible in each chapter and the chapter ending exercises Each chapter unfolds the structured systematic and patterned chemistry concepts in lucid and student friendly approach The book is without those unnecessary frills that make the bulk in other popular books in the market for the IITJEE An indispensable must have for in depth comprehension of Chemistry for the coveted IITJEE

Exel Withtm Objective Questions In Physical Chemistry S.K. Khanna; N.K. Verma; B. Kapila, **Educart CBSE Class 12 Physical Education Question Bank 2025-26 on new Syllabus 2026 (Includes Past Years Solved Questions)** Educart, 2025-05-28 **Physical Science - Chemistry Split with Online Learning Center Password Card (Chapters 1 And 8 - 13)** Bill W. Tillery, Tillery Bill, 2004-01-09 **The Journal of Physical Chemistry**, 1910 **The Journal of Physical Chemistry** Wilder Dwight Bancroft, 1910 Includes section New Books *Conceptual Physical Science* Paul G. Hewitt, 1998-11 *Conceptual Physical Science Third Edition* takes learning physical science to a new level by combining Hewitt s leading conceptual approach and friendly writing style in a new edition that provides stronger integration of the sciences more quantitative coverage and a wealth of new media resources to help readers The dynamic new media program includes hundreds of animations and interactive tutorials developed specifically for students taking physical science courses Media references throughout the book point readers to additional online help **KEY TOPICS** The book s consistent high quality coverage includes five new chapters on chemistry astronomy and earth science for an even more balanced approach to physical science For college instructors students or anyone interested in physical science **Kaplan Praxis 2005** Michael Palmer, Kaplan, Inc, 2004 From the most trusted name in the test preparation field Kaplan PRAXIS 2005 is an invaluable guide for succeeding on this difficult exam The Educational Review, 1986 **Chemistry** James E. Brady, Fred Senese, 2004-02-04 Publisher Description Vocabulary and Spelling Pergrande publishing company, Milwaukee, 1940 *Kaplan PRAXIS*, 2004 Kaplan PRAXIS. Kaplan, 2007 Taking prospective teachers step by step through the entire preparation process Kaplan PRAXIS 2007 Edition offers 2 full length PPST practice tests 1 full length PLT test and 3 Elementary Educations tests In depth review for PRAXIS I and II including the Pre Professional Skills Test PPST Multiple Subjects Assessment for Teachers MSAT Principles of Learning and Teaching PLT and Elementary Education tests Effective score raising strategies Question banks for English Science Mathematics and Social Studies Reference Manual of Government Positions Publishers for Exension Schools, inc., Milwaukee, 1939 *Biology/science Materials* Carolina Biological Supply Company, 1991 **Science Reporter**, 2007 **Reference Manual of Government Positions** Pergande

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Publisher's Monthly ,2003

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