

UNIT 10: PERIODIC TABLE **CHAPTER 10: PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------|
| _____ 18. A column in the periodic table | a. metals |
| _____ 19. A row in the periodic table | b. group |
| _____ 20. Group 1 elements | c. period |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 18 elements | e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- _____ 25. Group 1A elements except for hydrogen are located on the left side metals.
- _____ 26. Group 1A elements are the alkali metal metals.
- _____ 27. Group 1A elements are highly reactive nonmetals known as halogens.
- _____ 28. Group 1A elements are (not) nonmetallic elements located on nonmetallic metals.
- _____ 29. Alkali metals have properties of both metals and more nonmetallic metals.

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Michael Seilmaier



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McGraw-Hill Staff, 2001-07 *For All Practical Purposes* Consortium for Mathematics and Its Applications (U.S.), 2003 The sixth edition of the acclaimed classroom favorite offer a number of new features to help instructors strengthen the mathematical literacy of their students **Organic Chemistry** Francis A. Carey, 2000 A Market Leading Traditional Approach to Organic Chemistry Throughout all seven editions Organic Chemistry has been designed to meet the needs of the mainstream two semester undergraduate organic chemistry course This best selling text gives students a solid understanding of organic chemistry by stressing how fundamental reaction mechanisms function and reactions occur With the addition of handwritten solutions new cutting edge molecular illustrations updated spectroscopy coverage seamless integration of molecular modeling exercises and state of the art multimedia tools the 7th edition of Organic Chemistry clearly offers the most up to date approach to the study of organic chemistry *The Go-To Guide for Engineering Curricula, Grades 9-12* Cary I. Sneider, 2014-12-05 How to engineer change in your high school science classroom With the Next Generation Science Standards your students won't just be scientists they'll be engineers But you don't need to reinvent the wheel Seamlessly weave engineering and technology concepts into your high school math and science lessons with this collection of time tested engineering curricula for science classrooms Features include A handy table that leads you straight to the chapters you need In depth commentaries and illustrative examples A vivid picture of each curriculum its learning goals and how it addresses the NGSS More information on the integration of engineering and technology into high school science education

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