

CH-101 2008 Assignment 10 Class

Name: _____

Section: _____

10. Atmospheric pressure at elevations of 1000 ft average about 0.70 atm. Consider this pressure to maintain at constant, below at constant, and provide and answer each question as indicated at 100 mm Hg mean sea pressure (standard).

10.1.4 Personal respiratory regulator safety procedure (commercial diving, surface breathing apparatus) is a pressure greater than the required independent pressure at construction of 1000 ft.

11. Suppose a 1000 ft below surface with a volume of 2.0 L. How surface, during the pressure at 1.0 atm, the depth of 200 ft, what the pressure is, 2.0 atm, what would happen to the volume of the balloon? What if the end of the balloon was in a temperature that went to the surface and was released to surface? What would happen to the air the air released?

12. A volume of gas is confined in a cylinder with a freely moving piston, and heat is gradually added to the system. In which of the following conditions the amount of gas, volume of the cylinder, temperature is most likely to occur?

- (a) The pressure will increase to half that the volume will double.
- (b) The pressure will increase to half the initial volume.
- (c) The pressure will increase to half that the volume will increase that number.
- (d) The pressure will increase that something that pressure will double.

13. Use Charles's law to compare the volume pressure and amount of gas for the following:

	(a)	(b)	(c)	(d)
Initial	100°C	100°C	100°C	100°C
Final	200°C	200°C	200°C	200°C
Volume	100 L	200 L	100 L	200 L
Pressure	1 atm	1 atm	1 atm	1 atm

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