

# Atoms, Ions, & Molecules

	Atom	Ion	Molecule
Number of particles	 One 	One or more 	Two or more 
Charge	Neutral (equal protons and electrons)	Positive (lost electrons) negative (gained electrons)	Neutral (electrons shared between atoms)
Formation	Cannot be further broken down without losing chemical identity.	Formed by sharing electrons	Formed by sharing electrons between atoms.
Examples	Helium (He), Oxygen (O), Iron (Fe)	Sodium ion (Na <sup>+</sup> ), Chloride ion (Cl <sup>-</sup> ), Sulfate ion (SO <sub>4</sub> <sup>2-</sup> )	Water (H <sub>2</sub> O), Carbon dioxide (CO <sub>2</sub> ), Sugar (C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> )



# Chapter 7 Compounds Ions And Molecules

**AN Whitehead**



## **Chapter 7 Compounds Ions And Molecules:**

## Enjoying the Beat of Phrase: An Mental Symphony within **Chapter 7 Compounds Ions And Molecules**

In some sort of consumed by monitors and the ceaseless chatter of fast communication, the melodic beauty and mental symphony created by the prepared word usually fade in to the backdrop, eclipsed by the relentless sound and distractions that permeate our lives. But, situated within the pages of **Chapter 7 Compounds Ions And Molecules** a stunning fictional prize overflowing with organic thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, that interesting masterpiece conducts readers on a psychological trip, skillfully unraveling the hidden songs and profound affect resonating within each cautiously constructed phrase. Within the depths of this poignant analysis, we can examine the book is key harmonies, analyze its enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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