

CHAPTER 6

Section 3: Water and Solutions

Study Guide

In your textbook, read about water's polarity.

Label the diagram. Use these choices:

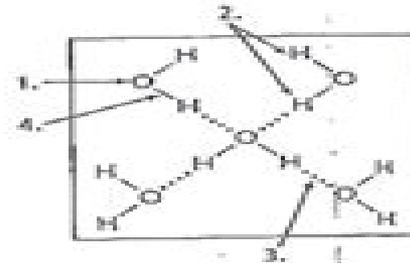
covalent bond

hydrogen bond

slightly negative end

slightly positive end

1. _____
2. _____
3. _____
4. _____



In your textbook, read about mixtures with water.

For each statement below, write true or false.

- _____ 5. A mixture is a combination of two or more substances in which each substance retains its individual characteristics.
- _____ 6. A suspension is a mixture that has a uniform composition throughout.
- _____ 7. In a mixture, the solvent is the substance that is dissolved.
- _____ 8. A mixture of sand and water is a heterogeneous mixture.
- _____ 9. A suspension is a homogeneous mixture in which water is mixed with a substance that does not dissolve in it.

In your textbook, read about acids and bases.

Use each of the terms below only once to complete the passage.

acids

bases

biology

buffers

hydrogen ions

neutral

pH

Substances that release hydrogen ions when dissolved in water are called

(10) _____. The more (11) _____ a substance

releases, the more acidic the solution becomes. Substances that release hydroxide ions when dissolved in

water are called (12) _____. Acids and bases are key substances in

(13) _____. The concentration of hydrogen ions in a solution is called

(14) _____. Pure water is (15) _____ and has a pH

value of 7.0. (16) _____ are weak acids or weak bases that can react with strong

acids or strong bases to keep the pH within a particular range.

Study Guide Chapter 6 Section 3 Water And Solutions

M Lipman



Study Guide Chapter 6 Section 3 Water And Solutions:

Study Guide for Atoms, Molecules, and Life Michael S. Matta, 1981 Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text **Excel Science Study Guide, Years 9-10** Will Marchmont, 2004 The book contains coverage of five major topic areas in the NSW School Certificate test Energy Force and Motion Atoms Elements and Compounds Structure and Function of Living Things Earth and Space Ecosystems Resources and Technology a chapter on Investigations and Problem Solving in Science to help with practical skills revision questions and chapter tests to help you remember important information a glossary and summary in each section of the book diagrams and illustrations to help your understanding a section to help you prepare for the School Certificate test a sample School Certificate test paper with answers answers to all questions **Environmental Science** Michael L. McKinney, Robert M. Schoch, Logan Yonavjak, 2012 Updated with the latest data from the field Environmental Science Systems and Solutions Fifth Edition explains the concepts and teaches the skills needed to understand multifaceted and often very complex environmental issues The authors present the arguments rebuttals evidence and counterevidence from many sides of the debate The Fifth Edition includes new Science in Action boxes which feature cutting edge case studies and essays contributed by subject matter experts that highlight recent and ongoing research within environmental science With an Earth as a system approach the text continues to emphasize Earth's intricate web of interactions among the biosphere atmosphere hydrosphere and lithosphere and how we are central components in these four spheres This flexible unbiased approach highlights 1 how matter cycles over time through Earth's systems 2 the importance of the input throughput output processes that describe the global environment 3 how human activities and consumption modify Earth's systems 4 and the scientific economic and policy solutions to environmental problems Revised and updated to reflect current trends and statistics within Environmental Science New content on renewable energy solar panels and compact fluorescent light bulbs The latest information on Hydropower and the advantages and disadvantages of hydroelectric energy The companion website includes robust learning tools that enable students to make full use of today's learning technology Students will find practice quizzes virtual flashcards answers to in text questions and links to additional coverage regarding material discussed in the text Instructor Resources include an instructor's manual Test Bank PowerPoint Lecture Outline Slides and a PowerPoint Image Bank **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a

summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Water Engineering Nazih K. Shamma, Lawrence K. Wang, 2015-05-27 Details the design and process of water supply systems tracing the progression from source to sink Organized and logical flow tracing the connections in the water supply system from the water's source to its eventual use Emphasized coverage of water supply infrastructure and the design of water treatment processes Inclusion of fundamentals and practical examples so as to connect theory with the realities of design Provision of useful reference for practicing engineers who require a more in depth coverage higher level students studying drinking water systems as well as students in preparation for the FE PE examinations Inclusion of examples and homework questions in both SI and US units

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