

# CHEMISTRY



**Atomic Structure**

Atoms are the smallest particles which cannot be created or destroyed in a chemical reaction. They are made up of three sub-particles: electrons, protons and neutrons.

**Atomic Number**

The number of protons present in the nucleus of an atom is called its atomic number. It is denoted by the symbol 'Z'.

**Atomic Mass**

The mass of an atom is called its atomic mass. It is denoted by the symbol 'A'.

**Isotopes**

Isotopes are atoms of the same element having the same atomic number but different mass numbers.

**Atomic Structure**

Atoms are made up of three sub-particles: electrons, protons and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.

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## Atomic Structure

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Element	Atomic Number (Z)	Atomic Mass (A)
Hydrogen	1	1
Helium	2	4
Lithium	3	7
Boron	5	11
Carbon	6	12
Nitrogen	7	14
Oxygen	8	16
Fluorine	9	19
Neon	10	20

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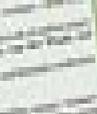
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**BarCharts, Inc, Lauren Scerbo**



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