

## Biology Chapter 2 Study Guide

1. What is the basic unit of all matter?
2. Name the three subatomic particles and give the mass, charge and location in the atom of each one.
3. What is the atomic number?
4. What is the atomic mass?
5. Draw elements 1-10 on the Periodic Table using the example on page 35. Be sure to show the electrons in their proper energy levels. Hint: Your Periodic Table App gives this information.
6. Define isotope.
7. Since different isotopes of atoms exist, explain how this leads to an average atomic mass for each element.
8. The isotopes of Carbon are radioactive, which means they decay at a certain known rate. How can knowing the amount of each isotope of Carbon in a shell that you find at the beach be utilized?
9. What is a compound or molecule?
10. Why do atoms form compounds or bond?
11. Describe ionic bonds and give an example of an ionic compound.
12. Describe a covalent bond and give an example of a covalent compound.
13. Why is water such an important compound?
14. Draw a molecule of water.
15. A water molecule is polar due to the way the atoms in water bond together. What kind of bond can form between water molecules due to this polarity?
16. Describe the difference between a mixture, solution and suspension.
17. What is an acid and how do they form?
18. What is a base and how does it form?
19. Write the chemical equation to show the disassociation of water in a solution to hydrogen and hydroxide ions.
20. Draw the pH scale. Show the values that represent a base, an acid and neutral solution and give one example of each.
21. Name the six most important atoms found in living things.
22. Why is carbon such an important atom in organic molecules? Discuss how it bonds and describe the types of molecules it can form.
23. How many covalent bonds are there between the carbon atoms in acetylene?
24. List the four major carbon containing compounds found in living things.
25. What are carbohydrates and why do living things need them? Distinguish between monosaccharides, disaccharides and polysaccharides.
26. What is a chemical reaction?
27. Write a chemical equation showing the reaction that occurs between hydrogen and oxygen to form water. Label the products and reactants.

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**AW Rasmussen**



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