

Experiment 11A: The Periodic Table

Procedure

Part 1 Electronic Configuration and the Periodic Table

1. Obtain a sheet of paper.

Part 2 Atomic Number, Atomic Radius, Ionization Energy and the Periodic Table

1. Obtain a sheet of paper (graph paper is preferred).

Part 3 Electronegativity and the Periodic Table

1. Obtain a sheet of paper.

Background

1. Examine the periodic table of elements (Figure 11.1). What relationship can be seen between an element's placement in the periodic table and its atomic configuration?

Elements in the same column (vertical column) of the periodic table have similar chemical properties. Though there are exceptions, most periodic trends are similar.

2. Examine the graph of atomic radii (plotted against increasing atomic number) (Figure 11.2). What is particularly noteworthy about it? Is it similar to the periodic trend?

Atomic radii tend to increase from left to right across a period and decrease from top to bottom in a group. The atomic radii of the noble gases are not included because they do not have a fixed radius. However, the general trend is that atomic radii increase from left to right across a period and decrease from top to bottom in a group.

3. Which group of elements has the largest atoms and which group has the smallest?

Examine the graph of ionization energy (plotted against increasing atomic number) (Figure 11.3). What is particularly noteworthy about it? Is it similar to the periodic trend?

4. Examine the graph of electronegativity (plotted against increasing atomic number) (Figure 11.4). What is particularly noteworthy about it? Is it similar to the periodic trend?

Electronegativity tends to increase from left to right across a period and decrease from top to bottom in a group. The noble gases are not included because they do not have a fixed electronegativity. However, the general trend is that electronegativity increases from left to right across a period and decreases from top to bottom in a group.

Laboratory Experiment 11a Answers Periodic Table

Xiaolong Qi



Laboratory Experiment 11a Answers Periodic Table:

The Enigmatic Realm of **Laboratory Experiment 11a Answers Periodic Table**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Laboratory Experiment 11a Answers Periodic Table** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of people who partake in its reading experience.

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