

**UNIT 10: PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

**Section 10.1** *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____
_____

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                                  |
|---|----------------------------------|
| _____ 18. A column in the periodic table                  | _____ a. metals                  |
| _____ 19. A row in the periodic table                     | _____ b. group                   |
| _____ 20. Group 1 elements                                | _____ c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements                                | _____ e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 8A elements are (or) noble gases elements known as inert gases.
- \_\_\_\_\_ 29. Alkaloids have properties of both metals and more reactive metals.

# Chapter 7 Study For Content Mastery Answer Key

**Minjie Lin**



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## **Table of Contents Chapter 7 Study For Content Mastery Answer Key**

1. Understanding the eBook Chapter 7 Study For Content Mastery Answer Key
  - The Rise of Digital Reading Chapter 7 Study For Content Mastery Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Study For Content Mastery Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 7 Study For Content Mastery Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Study For Content Mastery Answer Key
  - Personalized Recommendations
  - Chapter 7 Study For Content Mastery Answer Key User Reviews and Ratings
  - Chapter 7 Study For Content Mastery Answer Key and Bestseller Lists

5. Accessing Chapter 7 Study For Content Mastery Answer Key Free and Paid eBooks
  - Chapter 7 Study For Content Mastery Answer Key Public Domain eBooks
  - Chapter 7 Study For Content Mastery Answer Key eBook Subscription Services
  - Chapter 7 Study For Content Mastery Answer Key Budget-Friendly Options
6. Navigating Chapter 7 Study For Content Mastery Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 7 Study For Content Mastery Answer Key Compatibility with Devices
  - Chapter 7 Study For Content Mastery Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 7 Study For Content Mastery Answer Key
  - Highlighting and Note-Taking Chapter 7 Study For Content Mastery Answer Key
  - Interactive Elements Chapter 7 Study For Content Mastery Answer Key
8. Staying Engaged with Chapter 7 Study For Content Mastery Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 7 Study For Content Mastery Answer Key
9. Balancing eBooks and Physical Books Chapter 7 Study For Content Mastery Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 7 Study For Content Mastery Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter 7 Study For Content Mastery Answer Key
  - Setting Reading Goals Chapter 7 Study For Content Mastery Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 7 Study For Content Mastery Answer Key
  - Fact-Checking eBook Content of Chapter 7 Study For Content Mastery Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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