



**Directions:** Use the words listed below to complete the paragraphs. Some words will be used more than once.

	<b>solute</b>		<b>solvent</b>		<b>increases</b>
<b>decreases</b>		<b>temperature</b>		<b>nonpolar</b>	<b>polar</b>
<b>electrolyte</b>		<b>nonelectrolyte</b>		<b>higher</b>	<b>lower</b>

In a solution, the 1. \_\_\_\_\_ does the dissolving and the 2. \_\_\_\_\_ is the substance being dissolved. For solids dissolving in liquids, the speed of dissolving 3. \_\_\_\_\_ with increased surface area, with increased temperature, and with stirring. For gases in liquids, the speed of solubility 4. \_\_\_\_\_ with increased temperature but 5. \_\_\_\_\_ with increased pressure.

In general, polar solutes dissolve in polar solvents and do not dissolve in 6. \_\_\_\_\_ solvents. Usually, nonpolar solvents dissolve nonpolar solutes and do not dissolve 7. \_\_\_\_\_ solutes.

Solubility is usually expressed as the maximum number of grams of 8. \_\_\_\_\_ that will dissolve in 100 grams of solvent at a certain 9. \_\_\_\_\_.

A substance that forms charged ions in solution and can conduct electricity is called a(n) 10. \_\_\_\_\_. A(n) 11. \_\_\_\_\_ does not ionize in water and cannot conduct electricity. The boiling point of a water solution of any substance is 12. \_\_\_\_\_ than the boiling point of water. The freezing point of a water solution of any substance is 13. \_\_\_\_\_ than the freezing point of water.

# Directed For Content Mastery Physical Science Answers

**Huangqi Zhang**



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