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web assuming the volume of this mixture is 0 003 m³ kg of ceramic catalyst approximated as 1 kg of ceramic per 2 kg of solution thus 3 kg kg of ceramic the reactor volume is $v_r = 12 \text{ blocks} \times 5 \text{ kg ceramic block} = 0.003 \text{ m}^3 \text{ kg ceramic} \times 4 \text{h} = 0.72 \text{ m}^3$ answer problem 13 12 continued 1 d the reactor must contain a hot solution of sodium

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