

## Periodic Trends Lab

Periodic law states that when the elements are arranged in order of increasing atomic number, there is a periodic repetition of their chemical and physical properties. Elements within the same group have the most similar chemical and physical properties, although not identical. Elements in neighboring groups may also have similar chemical and physical properties, but to a lesser degree. Trends in properties occur within groups and across periods.

**PURPOSE:** In this activity, you will

- Interpret group and periodic trends in atomic radius, electronegativity, and ionization energy.
- Graph these trends.

**PROCEDURE:**

1. Complete the pre-lab section.
2. Using the data from **Table S** of your reference tables, prepare the graphs listed below. Staple your graphs to the back of this lab packet.

*On each graph be sure to:*

- USE PENCIL!
- Put an appropriate title and labels for each axis.
- Omit the transition elements.
- Connect the points you plot.
- Connect the points representing Group 1 elements with a green line.
- Connect the points representing Group 2 elements with a blue line.
- Connect the points representing Group 17 elements with a red line.
- Draw a triangle around the points representing the elements of Period 3.

GRAPH 1. Atomic radius vs. atomic number for elements 1-54.

GRAPH 2. Electronegativity vs. atomic number for elements 1-54.

GRAPH 3. Ionization energy vs. atomic number for elements 1-54.

3. Complete the conclusion section.

# Periodic Trends Lab Question Answers

**Kerry K. Karukstis, Timothy E. Elgren**



## Periodic Trends Lab Question Answers:

*Hands-On General Science Activities With Real-Life Applications* Pam Walker, Elaine Wood, 2008-04-21 In this second edition of *Hands On General Science Activities with Real Life Applications* Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5-12. The book offers a dynamic collection of classroom ready lessons, projects, and lab activities that encourage students to integrate basic science concepts and skills into everyday life.

[Exploring General Chemistry in the Laboratory](#) Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course. The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life. This lab manual covers topics such as composition of compounds, reactivity, stoichiometry, limiting reactants, gas laws, calorimetry, periodic trends, molecular structure, spectroscopy, kinetics, equilibria, thermodynamics, electrochemistry, intermolecular forces, solutions, and coordination complexes. By the end of this course, you should have a solid understanding of the basic concepts of chemistry, which will give you confidence as you embark on your career in science.

**O-level Chemistry Challenging Drill Questions (Concise) (Yellowreef)** Thomas Bond, Chris Hughes, 2013-11-07 questions from top schools since 2003, complete encyclopedia of over 350 questions, complete answer keys, full set of step-by-step solution approaches, available separately, complete edition and concise edition, eBooks available.

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**Prentice Hall Chemistry**, 2000 [Merrill Chemistry](#) Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *The Good Student*

Charles Jay Rop, 1996 **Chemistry 2007** David Wilson, 2006-12 Kaplan's guide includes 2 full-length practice tests.

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Developing and Sustaining a Research-supportive Curriculum Kerry K. Karukstis,Timothy E. Elgren,2007 This compendium of successful curricular and institutional practices to develop critical research skills emphasized the importance of the collective efforts of the undergraduate community to integrate research and education By collecting and disseminating a variety of mechanisms that are effective means of creating a research supportive undergraduate curriculum the Council on Undergraduate Research aims to encourage faculty and institutions to continue to seek creative useful and significant ways to promote learning through research Publisher's description Linne & Ringsrud's Clinical Laboratory Science - E-Book Mary Louise Turgeon,2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts **The World of Chemistry** Jones,Jones Staff,1996

**AP Chemistry For Dummies** Peter J. Mikulecky,Michelle Rose Gilman,Kate Brutlag,2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You'll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You'll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you'll

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