

Periodic Trends

ATOMIC RADIUS

1. What trend in atomic radius do you see as you go down a group/family on the periodic table?
2. What causes this trend?
3. What trend in atomic radius do you see as you go across a period/row on the periodic table?
4. What causes this trend?
5. Circle the atom in each pair that has the largest atomic radius.
 - a) Al B
 - b) S O
 - c) Br Cl
 - d) Na Al
 - e) O F
 - f) Mg Ca
6. Put the following elements in order from smallest to largest atomic radius **and** explain why:
C, O, Sn, Sr.

ELECTRONEGATIVITY

7. Define electronegativity
8. How does the ionic radius of a nonmetal compare with its atomic radius?
9. What trend in electronegativity do you see as you go down a group/family on the periodic table?
10. What causes this trend?
11. What trend in electronegativity do you see as you go across a period/row on the periodic table?
12. What causes this trend?
13. Circle the atom in each pair that has the greater electronegativity.
 - a) Ca Ga
 - b) Li O
 - c) Cl S
 - d) Br As
 - e) Ba Sr
 - f) O S

GENERAL QUESTIONS

14. Which group tends to form +1 ions? _____
15. Which group tends to form +2 ions? _____
16. Which group tends to form -1 ions? _____
17. Which group tends not to form ions or react? _____

Periodic Trends Reactivity Lab Answer Key

Rachel Sandford



Periodic Trends Reactivity Lab Answer Key:

Handbook of Formative Assessment Heidi Andrade, Gregory J. Cizek, 2010-04-02 Formative assessment has recently become a focus of renewed research as state and federal policy makers realize that summative assessments have reached a point of diminishing returns as a tool for increasing student achievement. Consequently, supporters of large scale testing programs are now beginning to consider the potential of formative assessments to improve student achievement. The mission of this handbook is to comprehensively profile this burgeoning field of study. Written by leading international scholars and practitioners, each chapter includes a discussion of key issues that dominate formative assessment policy and practice today as well as those that are likely to affect research and practice in the coming years. Key features include: Comprehensive: nineteen chapters cover all aspects of formative assessment including classroom assessment, large scale applications, technological applications, applications for special needs students, K-12 and post-secondary applications, psychometric considerations, case studies, and discussion of alternative assessment formats such as portfolios and performance assessments. Integrative: thoughtful attention is given to the integration of large scale and classroom assessments. Practical: provides practical guidance on how to conduct formative assessments that generate credible information to guide instruction. Global: provides perspectives from leading international scholars and practitioners whose expertise spans diverse settings, student populations, and educational systems. Accessible Style: although grounded in the latest research, the book's style and tone has been carefully crafted to make it accessible to both the textbook and professional markets. It will also be a critical reference book for researchers in teacher preparation, educational administration, and educational policy studies.

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Periodic Trends Reactivity Lab Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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