

## Unit Two Chemistry Exam

### Multiple Choice 20 marks

You need to record your answers in the appropriate section in the ANSWER BOOKLET.

#### Question 1

When ethanol dissolves in water the bonds formed between the ethanol and water molecules are known as

- A. dispersion forces
- B. ion-dipole attraction
- C. hydrogen bonds
- D. ionisation

#### Question 2

When NaOH dissociates in water, the equation that describes the process is

- A.  $\text{NaOH (s)} \rightarrow \text{Na}^+ \text{(aq)} + \text{OH}^- \text{(aq)}$
- B.  $\text{NaOH (s)} + \text{H}_2\text{O (l)} \rightarrow \text{NaO}^- \text{(aq)} + \text{H}_2\text{O}^+ \text{(aq)}$
- C.  $\text{NaOH (s)} + \text{H}_2\text{O (l)} \rightarrow \text{NaOH}_2^+ \text{(aq)} + \text{OH}^- \text{(aq)}$
- D.  $\text{NaOH (aq)} + \text{HCl (aq)} \rightarrow \text{NaCl (s)} + \text{H}_2\text{O (l)}$

#### Question 3

How much heat energy is required to raise the temperature of 250 mL of water from 17 °C to 100 °C? Assume that 1 mL of water has a mass of 1 g and that the specific heat capacity of water is  $4.2 \text{ J g}^{-1} \text{ }^\circ\text{C}^{-1}$ .

- A. 8.7 J
- B. 87,150 KJ
- C. 8.7 KJ
- D.  $8.7 \times 10^4 \text{ J}$

#### Question 4

A solution containing 0.25 g of NaCl dissolved in 50 mL of deionised water will have a concentration of

- A.  $5.0 \text{ gL}^{-1}$
- B. 4.3 M
- C.  $5.0 \times 10^{-3} \text{ mgL}^{-1}$
- D.  $2.5 \times 10^{-4} \text{ molL}$

#### Question 5

A solution of  $\text{CuSO}_4$  has a concentration of 2.50 M. This concentration is equivalent to

- A. 399 %m/v
- B. 39.9 %m/v
- C. 0.250 %m/v
- D. 25.0 m/v

# Chemistry Unit 2 Test Answers

**RS Peters**



## **Chemistry Unit 2 Test Answers:**

## **Chemistry Unit 2 Test Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Unit 2 Test Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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