

Chemistry Final Exam Study Guide 2021-22

The final exam will cover chapters 6, 7, 8 and 9, with a few very basic questions from chapters 10 and 11. Recommended ways to prepare for final exam.

- **review objectives, notes, worksheets, charts, quizzes and exams.**
- **make your own study guide**
- if you missed questions on test or quizzes, reread those topics in the textbook
- practice problems from each chapter. Appendix D in Pearson Chemistry has a problem bank with many of the answers included in the appendix. Use these for practice
- use quizzes for reviewing vocabulary, naming compounds, or other topics, most chapters of Pearson have quizzes already made or you can make your own... I have reviewed the quizzes listed on your powerschool pages and listed only those I felt were accurate.
- Study buddies, quizzing classmates or other students taking chemistry
- Take the standardized test in the textbook at the end of each chapter. I will post the answers to each of these

The exam will consist of 65-80 questions, all multiple choice. There will be copies of the periodic table, polyatomic ion chart, and transition metal charts attached to the back of the exam. You will need your calculator to complete the final exam.

Chapter 6: The Periodic Table (about 15 questions)

Vocabulary

proton
electron
atomic number
period
group
metal
nonmetal
representative element (Group A)
transition metal (Group B)
halogens
noble gas

Objectives (see p. 185 6.1 and 6.2)

- Know how the periodic table is arranged.
- Be able to use the periodic table to locate alkali metals, Group A elements, Group B elements and periods.
- Differentiate between metals, metalloids, and nonmetals.
- Be able to explain why noble gases can exist as single atoms, and usually don't react with other elements.
- **Know how to determine the number of valence electrons in a Group A element.**

Chemistry Final Exam Study Guide Packet

Becker, Charles F. Batten



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Chemistry George M. Bodner,Harry L. Pardue,1989-01-23 A modern experimental approach to first year chemistry This unique introductory account employs experimental observations to construct the principles of general chemistry An early introduction to observable descriptive chemistry lays the basis for the well developed exposition that follows

Assorted Practice Problems to Accompany Becker/Wentworth General Chemistry, Second Edition Becker,Charles F. Batten,1980

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Table of Contents Chemistry Final Exam Study Guide Packet

1. Understanding the eBook Chemistry Final Exam Study Guide Packet
 - The Rise of Digital Reading Chemistry Final Exam Study Guide Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Final Exam Study Guide Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Final Exam Study Guide Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Final Exam Study Guide Packet
 - Personalized Recommendations
 - Chemistry Final Exam Study Guide Packet User Reviews and Ratings
 - Chemistry Final Exam Study Guide Packet and Bestseller Lists
5. Accessing Chemistry Final Exam Study Guide Packet Free and Paid eBooks
 - Chemistry Final Exam Study Guide Packet Public Domain eBooks
 - Chemistry Final Exam Study Guide Packet eBook Subscription Services
 - Chemistry Final Exam Study Guide Packet Budget-Friendly Options

6. Navigating Chemistry Final Exam Study Guide Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Final Exam Study Guide Packet Compatibility with Devices
 - Chemistry Final Exam Study Guide Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Final Exam Study Guide Packet
 - Highlighting and Note-Taking Chemistry Final Exam Study Guide Packet
 - Interactive Elements Chemistry Final Exam Study Guide Packet
8. Staying Engaged with Chemistry Final Exam Study Guide Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Final Exam Study Guide Packet
9. Balancing eBooks and Physical Books Chemistry Final Exam Study Guide Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Final Exam Study Guide Packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Final Exam Study Guide Packet
 - Setting Reading Goals Chemistry Final Exam Study Guide Packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Final Exam Study Guide Packet
 - Fact-Checking eBook Content of Chemistry Final Exam Study Guide Packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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