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Chapter 11 - States of Matter: Liquids and Solids

1. Which of the following statements concerning liquids is incorrect?

- A) The volume of a liquid changes very little with pressure.
- B) Liquids are relatively incompressible.
- C) Liquid molecules move slowly compared to solids.
- D) Non-volatile liquids have low vapor pressures at room temperature.
- E) The molecules of a liquid are in constant random motion.

ANS: C

PTS: 1

DIF: easy

REF: 11.1

OBJ: Compare a gas, a liquid, and a solid using a kinetic-molecular theory description.

TOP: phases | liquid

KEY: properties of liquids

MSC: general chemistry

2. Which of the following statements concerning solids, liquids and gases is/are correct?

- 1. The close contact of the particles in a solid prevents all possible motion, including vibrational motion.
- 2. Liquids and gases are both considered fluids.
- 3. The postulate from kinetic-molecular theory that the particles in a gas are in constant random motion is equally applicable to liquids, except the particles are more tightly packed in a gas.

A) 1 only

B) 2 only

C) 3 only

D) 1 and 2

E) 2 and 3

ANS: B

PTS: 1

DIF: easy

REF: 11.1

OBJ: Compare a gas, a liquid, and a solid using a kinetic-molecular theory description.

TOP: phases | liquid

3. What is the name for the following phase change?

General Chemistry Ebbing 10th Edition Test Bank

Huangqi Zhang



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