

9/2/16

S. Sai Krishna

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ClassmateThermochemistry

$$Q = \Delta E - W$$

$$Q = \Delta E - (-P\Delta V)$$

$$Q = \Delta E + P\Delta V$$

- At constant Volume.

$$Q = \Delta E$$

- At const. Pressure.

$$Q_p = \Delta E + P\Delta V$$

$$H = \Delta E + P\Delta V$$

$$H = \Delta E + \Delta nRT$$

$\Delta n =$ no. of moles of Prod
- no. of moles of react.

→ For ΔH .

- If ΔH is -ve. reaction is exothermic.
- If ΔH is +ve. reaction is endothermic.

Q. The heat of combustion of ethane at 17°C and const volume is -322.1 kJ . Calculate the heat of combustion at constant pressure.



Be Engineering Chemistry Notes 2016

Michael Seilmaier



Be Engineering Chemistry Notes 2016:

The book delves into Be Engineering Chemistry Notes 2016. Be Engineering Chemistry Notes 2016 is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Be Engineering Chemistry Notes 2016, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Be Engineering Chemistry Notes 2016
- Chapter 2: Essential Elements of Be Engineering Chemistry Notes 2016
- Chapter 3: Be Engineering Chemistry Notes 2016 in Everyday Life
- Chapter 4: Be Engineering Chemistry Notes 2016 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Be Engineering Chemistry Notes 2016. The first chapter will explore what Be Engineering Chemistry Notes 2016 is, why Be Engineering Chemistry Notes 2016 is vital, and how to effectively learn about Be Engineering Chemistry Notes 2016.

3. In chapter 2, the author will delve into the foundational concepts of Be Engineering Chemistry Notes 2016. The second chapter will elucidate the essential principles that need to be understood to grasp Be Engineering Chemistry Notes 2016 in its entirety.

4. In chapter 3, this book will examine the practical applications of Be Engineering Chemistry Notes 2016 in daily life. The third chapter will showcase real-world examples of how Be Engineering Chemistry Notes 2016 can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Be Engineering Chemistry Notes 2016 in specific contexts. The fourth chapter will explore how Be Engineering Chemistry Notes 2016 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Be Engineering Chemistry Notes 2016. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Be Engineering Chemistry Notes 2016.

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