

# KEY TERMS: Mix and Match

Instructions: Match the term on the left with the correct definition on the right. Write the letter of the correct definition in the space provided.

1. <b>Hydrophilic</b>	21. <b>Hydrophobic</b> - does not interact with water.
2. <b>Hydrophobic</b>	22. <b>Hydrophilic</b> - interacts with water. Example: polar molecules.
3. <b>Amphipathic</b>	23. <b>Amphipathic</b> - has both hydrophilic and hydrophobic parts. Example: phospholipids.
4. <b>Water</b>	24. <b>Water</b> - a polar molecule with a high heat capacity and a high heat of vaporization.
5. <b>Ice</b>	25. <b>Ice</b> - a solid form of water with a lower density than liquid water.
6. <b>Hydrogen bond</b>	26. <b>Hydrogen bond</b> - a weak attractive force between polar molecules.
7. <b>Hydrophobic interaction</b>	27. <b>Hydrophobic interaction</b> - the tendency of non-polar molecules to cluster together in water.
8. <b>Hydrophilic interaction</b>	28. <b>Hydrophilic interaction</b> - the tendency of polar molecules to interact with water.
9. <b>Hydrophobic effect</b>	29. <b>Hydrophobic effect</b> - the tendency of non-polar molecules to cluster together in water.
10. <b>Hydrophilic effect</b>	30. <b>Hydrophilic effect</b> - the tendency of polar molecules to interact with water.
11. <b>Hydrophobic force</b>	31. <b>Hydrophobic force</b> - the force that causes non-polar molecules to cluster together in water.
12. <b>Hydrophilic force</b>	32. <b>Hydrophilic force</b> - the force that causes polar molecules to interact with water.
13. <b>Hydrophobic force</b>	33. <b>Hydrophobic force</b> - the force that causes non-polar molecules to cluster together in water.
14. <b>Hydrophilic force</b>	34. <b>Hydrophilic force</b> - the force that causes polar molecules to interact with water.
15. <b>Hydrophobic force</b>	35. <b>Hydrophobic force</b> - the force that causes non-polar molecules to cluster together in water.
16. <b>Hydrophilic force</b>	36. <b>Hydrophilic force</b> - the force that causes polar molecules to interact with water.
17. <b>Hydrophobic force</b>	37. <b>Hydrophobic force</b> - the force that causes non-polar molecules to cluster together in water.
18. <b>Hydrophilic force</b>	38. <b>Hydrophilic force</b> - the force that causes polar molecules to interact with water.
19. <b>Hydrophobic force</b>	39. <b>Hydrophobic force</b> - the force that causes non-polar molecules to cluster together in water.
20. <b>Hydrophilic force</b>	40. <b>Hydrophilic force</b> - the force that causes polar molecules to interact with water.

# Key Terms Mix And Match Biozone Answers

**Joacim Rocklöv**



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