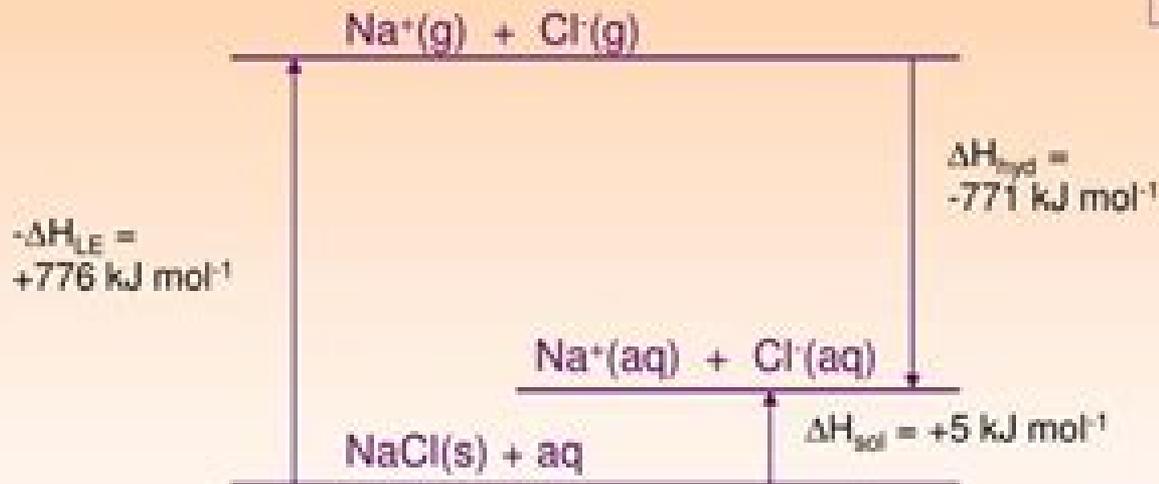


Enthalpy of Solution

- Enthalpy Diagram

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$$\Delta H_{\text{sol}} = -\Delta H_{\text{LE}} + (\Delta H_{\text{hyd}}(\text{cation}) + \Delta H_{\text{hyd}}(\text{anion}))$$

$$\Delta H_{\text{sol}} = -776 + -771 = +5 \text{ kJ mol}^{-1}$$

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