

Kali Linux CTF Blueprints

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Cameron Buchanan

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web a reaction and its experimental rate law are represented below $2 \text{ ICl} + \text{H}_2 \text{g} \rightarrow 2 \text{ HI} + 2 \text{ HCl} \text{g}$ rate $k [\text{ICl}]^2 [\text{H}_2]$ a chemist proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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web reaction rates google classroom $\text{mg s} + 2 \text{hcl a q} \rightarrow \text{mgcl}_2 + 2 \text{a q} + \text{h}_2 \text{g}$ the reaction between mg s and hcl a q is represented by the equation above in a kinetics experiment a 0.080 g sample of mg s ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M hcl a q at 25 °C

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