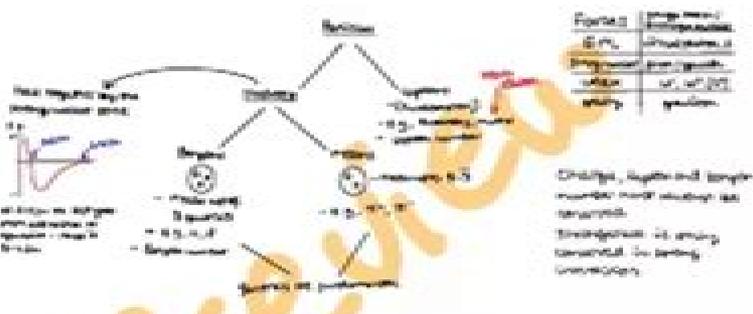


# Particles and Quantum



**Excitation:** when a particle and its corresponding antiparticle collide and their rest energy is converted into photons.



**Pair production:** when a photon turns into a particle and its corresponding antiparticle. The photon must have at least the sum amount of energy to create the particles.



## Excitation

Exciting means they give electrons enough energy to climb their atomic shells.

**Alpha particles** - Helium nucleus,  $^4_2\text{He}$

- Highly ionizing, so it is dangerous.
- Highly penetrating - stopped by a piece of paper or foil on their way.

**Beta** - fast moving electron  $\rightarrow e^-$

- Medium penetrating and velocity slowing - stopped by a few cm of aluminium.
- Beta decay -  $^0_{-1}\text{e} + ^0_0\gamma$  - stopped by a few mm of concrete.



**Gamma** - high energy. Can ray or photon emitted from a high energy nucleus  $\rightarrow \gamma$

- Highly ionizing.
- Highly penetrating - stopped by a few m of concrete.

**Isotopes** - same element - same number of protons

- different relative atomic mass - different number of neutrons

$^A_Z X$  where A is mass number, Z is atomic number.

**Nuclear charge** -  $\frac{Q}{m} [C/kg]$

- charge to mass ratio

**Electron volt** - charge of an electron when accelerated through a pot of one volt

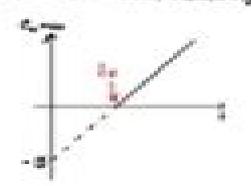
- $1.6 \times 10^{-19} J$
- eV



## Photoelectric effect

Use photons of sufficient energy on electrons by liberating from the surface of the metal, liberating them.

- $E_{\text{photon}} = hf = \phi + E_{\text{max}}$
- $\phi$  - photon absorbed -  $h\nu$  (work function)
- $E_{\text{max}}$  - frequency
- $\phi$  - work function - minimum energy to liberate electrons
- $E_{\text{max}}$  - energy that will give an  $e^-$  its maximum
- $E_{\text{max}}$  - stopped frequency - minimum frequency needed for electrons to be liberated



The photoelectric effect proved the particle nature of light due to the fact that the maximum kinetic energy of electrons  $E_{\text{max}}$  is not linearly dependent on the wavelength of light. Instead, it is linearly dependent on the frequency of light.  $E_{\text{max}}$  is zero for frequencies below the threshold frequency, and it increases linearly for frequencies above the threshold frequency.

**Stopping potential** - the pot. needed to stop any electrons from reaching the other side.

- $E_{\text{max}} = eV_s$
- $E_{\text{max}}$  - change in maximum
- $V_s$  - stopping potential

## De Broglie

The de Broglie wavelength is the wavelength that a particle can have. The wave nature of a particle will occur by being described as a packet of waves which propagate through space and time. This is because the electron is displaying wave-like behavior.



- $\lambda = \frac{h}{p}$
- $\lambda$  - de Broglie wavelength
- $p$  - momentum / mass x velocity

## Fluorescent tubes



These tubes contain mercury vapor. A high voltage is applied. This excites the gas, causing it to emit light. The light is then converted into visible light by the phosphor coating. The phosphor coating is a material that emits light when it is struck by high energy particles. The phosphor coating is a material that emits light when it is struck by high energy particles.

## Energy levels

Electrons can be excited to higher energy levels. Either by absorbing photons from the light or by the application of an electric field. When an electron returns to its ground state, it emits a photon of light.

**Ionization energy** - is enough energy to pull an electron out of an atom, so will completely change the atom.



# A Level Physics Revision Notes 2015 S Cool The Revision

**Didier Musso**



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