

8th Grade Science STAAR Review

Cheat Sheet

Matter & Energy

Subatomic Particles		
Protons	Neutrons	Electrons
Positive	Neutral	Negative
1 AMU	1 AMU	0 AMU
Nucleus	Nucleus	Electron Cloud

- Solids**
1. Order of particles decreases
 2. Interchemical and physical properties

- Liquids**
1. Order of particles decreases
 2. Interchemical and physical properties



Mg (Magnesium)

Atomic Number = 12

Protons = 12

Electrons = 12

Mass Number = 24

Atomic Number = 12

$24 - 12 = 12$ Neutrons

Magnesium is found in group 2, period 3 so that means it will have 2 Valence Electrons and 3 Electron Shells.

2 Valence Electrons will make it very reactive

Group 1, 6, 7 highest reactivity



Water Molecule (The small circle is the right of the element Oxygen, 2 circles of Oxygen or 2O atoms, all together that's 2O₂ that's why the coefficient was placed when they added to the water product)

Molecule Oxygen (circle) Group 16 Elements on the Periodic Table (Carbon, Hydrogen Group)

Molecule 2 (circles) as most of oxygen (the big number's coefficient) 2 molecule of O₂ and 2 molecule of H₂

Compound that is made of two different elements together Oxygen, H₂O and CO₂

Force, Motion, & Energy

Newton's 3 Laws of Motion

1. States the object at rest stays at rest and the object in motion stays in motion unless acted on by an unbalanced force.
2. Force & Acceleration: The more mass an object has the more force needed to accelerate it (F=MA)
3. Action-Reaction: For every action there is an equal and opposite reaction.



- Force
- Acceleration
- Reaction



- Force
- Force application
- Time to do double of the jump time

Speed - The distance and object travels in a certain amount of time.

Velocity - Speed in a certain direction.

Acceleration - When an object, speed up, slow down, or change direction. (A change in velocity)

Distance-Time



Velocity-Time



Metals - good conductors, solid, malleable, luster

Nonmetals - not good conductors, brittle or gases

Metalloids - share properties of both metals and nonmetals



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Albert A Gayle



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