

## Triple Integration with Maple

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Maple is helpful in setting up triple iterated integrals over regions in  $\mathcal{R}^3$ . Just as for double integrals, the key step is determining the limits of integration. That is easier if you imagine an arrow shooting through the region  $\mathcal{E}$  of integration parallel to the axis of the variable with respect to which the first integration takes place. Plotting the region  $\mathcal{E}$  reduces the vividness of imagination necessary to see the corresponding limits.

Fubini's theorem for triple integrals states that the value of a triple integral of a continuous function  $f$  over a region  $\mathcal{E}$  in  $\mathcal{R}^3$  is a triple iterated integral. Suppose that  $\mathcal{E}$  is a "Type 1" region between surfaces with equations  $z = h(x, y)$  and  $z = k(x, y)$  and has perpendicular projection  $D$  on the  $xy$ -plane. Then

$$\iiint_{\mathcal{E}} f(x, y, z) \, dz \, dy \, dx = \iint_D \int_{h(x, y)}^{k(x, y)} f(x, y, z) \, dz \, dy \, dx$$

The remaining double integral over the region  $D$  in the  $xy$ -plane is evaluated by iteration as before.

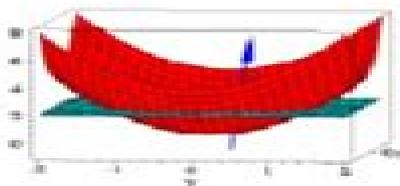
**Example 1.** Evaluate  $\iiint_{\mathcal{E}} z \, dz \, dy \, dx$ , where  $\mathcal{E}$  is the region between the paraboloid  $z = x^2 + y^2$  and the plane  $z = 2$ .

**Solution.** The following Maple routine plots the region  $\mathcal{E}$  with an upward-pointing arrow passing through it. The entry point of the arrow occurs on the surface whose equation  $z = h(x, y)$  gives the lower limit of the first integration, which will take place with respect to  $z$ . The exit point of the arrow is on the surface whose equation  $z = k(x, y)$  gives the upper limit of that first integration.

As for two-dimensional arrows, when working with other regions modify the parameters in the `ARROW` command to size the arrow appropriately. The first and second triples within the command specify the starting and end point of the arrow; the third one specifies a normal direction for the plane in which the arrow is seen; note that it does not appear perfectly vertical even though the direction vector  $\mathbf{k}$  was specified. Adjust the direction vector to get an image that you like.

```
> with(plots):
with(plottools):
regionbot := plot3d( x^2 + y^2, x = -2..2, y = -2..2, color = red );
regiontop := plot3d( 2, x = -2..2, y = -2..2, color = cyan );
regname := textplot3d([0.5, 0.5, 2.5, "E"], font = [TIMES, ITALIC, 14], color = black);
alice := arrow([3, .3, -1], [3, .3, 8.5], [0, 0, 1], 1, 4, .25, color = blue);
display(regionbot, regiontop, alice, regname, axes = boxed);
```

Warning: the name `changeprocs` has been redefined



Maple's real-time rotation feature reveals that the bottom boundary of the region is the paraboloid  $z = x^2 + y^2$ , and its top boundary is the plane  $z = 2$ . The lower limit of the first integration, with respect to  $z$ , corresponds to the equation of the lower surface, and the upper limit is  $z = 2$ . Hence,

$$\iiint_{\mathcal{E}} f(x, y, z) \, dz \, dy \, dx = \iint_D \int_{x^2 + y^2}^2 f(x, y, z) \, dz \, dx \, dy$$

# Triple Integration With Maple Uconn

**S Marginson**



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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is  $2\text{cl aq} + 2\text{ag aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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jun 23 2022 what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative

descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds  $\text{Cl}^-$  aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13  $2\text{Ag}^+ \text{aq} + 2\text{F}^- \text{aq} + 2\text{NH}_4^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7 \text{s} + 2\text{NH}_4 \text{aq} + 2\text{F}^- \text{aq}$  because two  $\text{NH}_4^+ \text{aq}$

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