

# AP Chemistry Final Study Guide

## Unit 1: Atomic Structure and Properties (7%–9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
  - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
  - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
  - The greater the charge of the nucleus, the more energy is required to ionize an electron
  - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
  - When atoms absorb energy, electrons jump to higher energy levels
  - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
  - The faster an ejected electron is going, the more kinetic energy it has
  - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
  - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
    - Nickel:  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$  (notice the d-block)
    - Nickel (with shorthand):  $[\text{Ar}]4s^2 3d^8$
- **Law of Definite Proportions:**
  - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
  - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
  - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

# Ap Chemistry Study Guide Amazon

**A Loxley**



**Ap Chemistry Study Guide Amazon:**

## Whispering the Secrets of Language: An Mental Journey through **Ap Chemistry Study Guide Amazon**

In a digitally-driven world where monitors reign great and instant conversation drowns out the subtleties of language, the profound strategies and mental subtleties concealed within phrases frequently go unheard. However, located within the pages of **Ap Chemistry Study Guide Amazon** a interesting fictional treasure sporting with raw feelings, lies an extraordinary quest waiting to be undertaken. Published by a skilled wordsmith, that marvelous opus attracts visitors on an introspective journey, softly unraveling the veiled truths and profound impact resonating within ab muscles cloth of each and every word. Within the emotional depths of this poignant evaluation, we can embark upon a honest exploration of the book is primary subjects, dissect their charming writing style, and yield to the effective resonance it evokes deep within the recesses of readers hearts.

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