

Answers to Chapter 3 Study Questions

1. a) $3(12.0) + 8(1.01) + 3(16.0) = 92.1 \text{ g/mole}$ b) 92.1 g c) 6.02×10^{23} molecules
 d) $0.217 \text{ moles} \times \frac{92.1 \text{ g}}{1 \text{ mole}} = 20.0 \text{ g}$
2. a) 4 atoms (one N + 3 H)
 b) $1 \text{ mole } \text{NH}_3 \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole } \text{NH}_3} \times \frac{4 \text{ atoms}}{1 \text{ molecule}} = 2.41 \times 10^{24} \text{ atoms}$
 c) $3.40 \text{ g } \text{NH}_3 \times \frac{1 \text{ mole } \text{NH}_3}{17.0 \text{ g } \text{NH}_3} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole } \text{NH}_3} \times \frac{4 \text{ atoms}}{1 \text{ molecule}} = 4.82 \times 10^{23} \text{ atoms}$
3. Molar mass of $\text{NaNO}_2 = 23.0 + 14.0 + 2(16.0) = 69.0 \text{ g/mole}$
 $\% \text{ Na} = 23.0/69.0 = 33.3\% \text{ Na}$; $\% \text{ N} = 14.0/69.0 = 20.3\% \text{ N}$; $\% \text{ O} = 2(16.0)/69.0 = 46.4\% \text{ O}$
33.3% Na, 20.3% N and 46.4% O.
4. a) In 100 g of this compound, there are 40.7 g C, 5.1 g H, and 54.2 g O
 $40.7 \text{ g C} \times \frac{1 \text{ mole C}}{12.0 \text{ g C}} = 3.39 \text{ moles C}$ $3.39/3.39 = 1 \times 2 = 2$
 $5.1 \text{ g H} \times \frac{1 \text{ mole H}}{1.0 \text{ g H}} = 5.1 \text{ moles H}$ $5.1/3.39 = 1.5 \times 2 = 3$
 $54.2 \text{ g O} \times \frac{1 \text{ mole O}}{16.0 \text{ g O}} = 3.39 \text{ moles O}$ $3.39/3.39 = 1 \times 2 = 2$
 empirical formula = **$\text{C}_2\text{H}_3\text{O}_2$**
- b) Molar mass of $\text{C}_2\text{H}_3\text{O}_2 = 2(12.0) + 3(1.0) + 2(16.0) = 59.0 \text{ g/mole}$
 $118/59.0 = 2 \rightarrow$ molecular formula = **$\text{C}_4\text{H}_6\text{O}_4$**
5. a) In 25.0 g of compound, there are 7.20 g Mg, 3.55 g C and $25.0 - (7.20 + 3.55) = 14.25 \text{ g O}$.
 $7.20 \text{ g Mg} \times \frac{1 \text{ mole Mg}}{24.3 \text{ g Mg}} = 0.296 \text{ moles Mg}$ $0.296/0.296 = 1$
 $3.55 \text{ g C} \times \frac{1 \text{ mole C}}{12.0 \text{ g C}} = 0.296 \text{ moles C}$ $0.296/0.296 = 1$
 $14.25 \text{ g O} \times \frac{1 \text{ mole O}}{16.0 \text{ g O}} = 0.891 \text{ moles O}$ $0.891/0.296 = 3$
 formula = **MgCO_3**
- b) $\% \text{ Mg} = 7.20/25.0 = 28.8\% \text{ Mg}$; $\% \text{ C} = 3.55/25.0 = 14.2\% \text{ C}$; $\% \text{ O} = 14.25/25.0 = 57.0\% \text{ O}$. (You should get the same result using molar mass and atomic masses.)
- c) $13.9 \text{ g compound} \times \frac{28.8 \text{ g Mg}}{100 \text{ g compound}} = 4.00 \text{ g Mg}$

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**Holt Rinehart & Winston, Holt,
Rinehart and Winston Staff**

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Fundamentals of Analytical Chemistry Douglas A. Skoog, 2004 This text is known for its readability combined with a systematic rigorous approach Extensive coverage of the principles and practices of quantitative chemistry ensures suitability for chemistry majors

Process Safety for Engineers CCPS (Center for Chemical Process Safety), 2022-04-12 Process Safety for Engineers Familiarizes an engineer new to process safety with the concept of process safety management In this significantly revised second edition of Process Safety for Engineers An Introduction CCPS delivers a comprehensive book showing how Process Safety concepts are used to reduce operational risks Students new engineers and others new to process safety will benefit from this book In this updated edition each chapter begins with a detailed incident case study provides steps that help address issues and contains problem sets which can be assigned to students The second edition covers Process Safety including an overview of CCPS Risk Based Process Safety Hazards specifically fire and explosion reactive chemical and toxicity Design considerations for hazard control including Hazard Identification and Risk Analysis Management of operational risk including management of change In addition the book presents how Process Safety performance is monitored and sustained The associated online resources are linked to the latest online CCPS resources and lectures

Journal of Biological Education, 1989

Developing Language and Literacy in English across the Secondary School Curriculum Urszula Clark, 2018-07-21 This book draws on original research and a language based pedagogy approach to examine how secondary schools in the UK can devise and implement coherent language and literacy across curriculum policies and strategies so that grammar and associated metalanguage becomes an integral part of their day to day curriculum practices The research was undertaken in three 11 to 18 secondary schools in England where the majority of students are categorised as having English as a second language EAL and where a significant minority are also socially disadvantaged in two of the three The author argues that paying explicit attention to the linguistic structures through which subject knowledge is realised can be of benefit to all pupils in ways that are also socially just and democratic This book provides an important bridge between academic theory and educational practice that will appeal to applied linguists and sociolinguists as well as to teachers teacher trainers and practitioners

Linne & Ringsrud's Clinical Laboratory Science - E-Book Mary Louise Turgeon, 2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts

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collaborative improvement of teaching and learning Includes US and international case studies powerful metaphors application exercises a leader s guide a companion website digital templates and more Learn what lesson study and collaborative inquiry can and should look like Find the guidance you need to lead and support schoolwide inquiry based improvement A true inspiration for educators who want to improve both their own craft and the methods of the profession

Jim Stigler James Hiebert Authors of *The Teaching Gap* **Books in Print Supplement**, 2002 **Exploring Earth and Space** Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter *Chemistry* John S. Phillips, Cheryl Wistrom, 2000 *Science Interactions*, 1996 **Cooperative Learning in the Chemistry Classroom** Melissa Ann Flynn, 1999 Teaching Undergraduate Science Linda C. Hodges, 2025-09-25 *Teaching Undergraduate Science A Guide to Overcoming Obstacles to Student Learning* offers college and university instructors evidence based strategies to help students learn those specific skills and habits of mind necessary for succeeding in STEM fields Updated and expanded from the first edition this text elaborates on critical factors in cultivating student success including how to engender a sense of belonging and agency in STEM engage students in their learning and foster deliberate practice Hodges provides frank guidance on the relative effort and outcomes for each strategy allowing instructors to choose techniques best suited to their aims and contexts While focusing primarily on face to face classes this resource also addresses how to work between online resources and physical spaces Hodges years of experience working as and with STEM faculty provides a personal connection to the research shared producing an accessible practical and enjoyable read Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Forthcoming Books* Rose Arny, 2003 **Glencoe Science** McGraw-Hill Staff, 2001-06 Teaching English-Medium Instruction Courses in Higher Education Ruth Breeze, Carmen Sancho Guinda, 2021-10-07 This book provides practical help and guidance for non native English speaking higher education lecturers faced with the need to deliver lectures and seminars in English It builds on the authors years of experience as researchers and teacher trainers in the area of English Medium Instruction EMI combining practical advice and research findings with useful case studies from different global settings including Australia China Hong Kong Slovakia Spain the UK and the USA and a range of subject areas such as philosophy mathematics and genetics The authors present an overview of what generally happens when university teachers make the transition to teaching in English After dispelling some common myths and setting out priorities Ruth Breeze and Carmen Sancho Guinda move on to explain how practitioners can prepare to give lectures and interact with both local and international students effectively in English tackling difficult issues such as encouraging participation promoting creativity and critical thinking and evaluating written student work The final chapters address good practices in EMI proposing ways to achieve excellence in global settings **Chemistry** William L. Masterton, 1996-08 This new edition of CHEMISTRY PRINCIPLES AND REACTIONS continues to provide students with the

core material essential to understanding the principles of general chemistry Masterton and Hurley cover the basics without sacrificing the essentials appealing to several markets Appropriate for either a one or two semester course CHEMISTRY PRINCIPLES AND REACTIONS Fifth Edition is three hundred pages shorter than most general chemistry texts and lives up to its long standing reputation as THE student oriented text Though this text is shorter in length than most other General Chemistry books it is not lower in level and with the addition of the large volume of content provided by the revolutionary GENERAL CHEMISTRY INTERACTIVE 3 0 CD ROM that is included with every copy it has a depth and breadth rivaling much longer books *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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