

CHAPTER 3 Matter—Properties and Changes

3.1 Properties of Matter

All of the material—the “stuff”—around us is matter. A **substance** is matter that has a uniform and unchanging composition. For example, water is a pure substance. No matter where it is found, a sample of water will have the same composition as any other sample of water.

A **physical property** of a substance is a characteristic that can be observed and measured without changing the composition of the substance. Words such as *hard*, *soft*, *shiny*, *dull*, *brittle*, *flexible*, *heavy* (in density), and *light* (in density) are used to describe physical properties.

A **chemical property** describes the ability of a substance to combine with or change into one or more other substances. For example, the ability of iron to form rust when combined with air is a chemical property of iron. The inability of a substance to combine with another substance is also a chemical property. For example, the inability to combine with most other substances is a chemical property of gold.

Practice Problems

1. Identify each of the following as an example of a physical property or a chemical property.
 - a. Silver tarnishes when it comes in contact with hydrogen sulfide in the air.
 - b. A sheet of copper can be pounded into a bowl.
 - c. Barium melts at 725°C .
 - d. Helium does not react with any other element.
 - e. A bar of lead is more easily bent than is a bar of aluminum of the same size.
 - f. Potassium metal is kept submerged in oil to prevent contact with oxygen or water.

Chapter 3 Matter Properties And Changes

Peipei Pang



Chapter 3 Matter Properties And Changes:

This book delves into Chapter 3 Matter Properties And Changes. Chapter 3 Matter Properties And Changes is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chapter 3 Matter Properties And Changes, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Chapter 3 Matter Properties And Changes
- Chapter 2: Essential Elements of Chapter 3 Matter Properties And Changes
- Chapter 3: Chapter 3 Matter Properties And Changes in Everyday Life
- Chapter 4: Chapter 3 Matter Properties And Changes in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Chapter 3 Matter Properties And Changes. The first chapter will explore what Chapter 3 Matter Properties And Changes is, why Chapter 3 Matter Properties And Changes is vital, and how to effectively learn about Chapter 3 Matter Properties And Changes.

3. In chapter 2, this book will delve into the foundational concepts of Chapter 3 Matter Properties And Changes. This chapter will elucidate the essential principles that need to be understood to grasp Chapter 3 Matter Properties And Changes in its entirety.

4. In chapter 3, this book will examine the practical applications of Chapter 3 Matter Properties And Changes in daily life. This chapter will showcase real-world examples of how Chapter 3 Matter Properties And Changes can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Chapter 3 Matter Properties And Changes in specific contexts. The fourth chapter will explore how Chapter 3 Matter Properties And Changes is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Chapter 3 Matter Properties And Changes. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 3 Matter Properties And Changes.

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