

UNIT 10: PERIODIC TABLE **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	10.
atomic number	10.
electron configuration	10.
chemical name	10.
chemical symbol	10.

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------|
| _____ 10. A column in the periodic table | a. metals |
| _____ 10. A row in the periodic table | b. group |
| _____ 20. Group 1 elements | c. period |
| _____ 20. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 20. Group 8 elements | e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 20. There are two representative elements of elements.
- _____ 20. Most elements besides boron of the elements in the periodic table are metalloids.
- _____ 20. Group 1A elements except for hydrogen are located on the left of metals.
- _____ 20. Group 2A elements are the alkaline earth metals.
- _____ 20. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 20. Group 8A elements are (a) unreactive elements known as noble gases.
- _____ 20. Metalloids have properties of both metals and nonmetals.

Chapter 7 Study Guide For Content Mastery Answer Key

Y Pai



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