

UNIT 10: PERIODIC TABLE **STANDARD CURRICULAR PERFORMANCE INDICATOR**

Section 1.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 16. A column in the periodic table | _____ a. metals |
| _____ 17. A row in the periodic table | _____ b. group |
| _____ 18. Group 1 elements | _____ c. period |
| _____ 19. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 20. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of platinum.
- _____ 22. Most elements besides boron of the elements in the periodic table are metalloids.
- _____ 23. Group 1A elements except for hydrogen are located on the left side.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are (a) unreactive elements known as noble gases.
- _____ 27. Metalloids have properties of both metals and nonmetals.

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Michal Rosen-Zvi



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