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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{CrO}_7^{2-} \text{aq} \rightarrow \text{Ag}_2\text{CrO}_7 \text{s}$

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chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change
oxidation number or composition during a reaction products

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