

O level Chemistry Notes



Atoms: Building block of matter. It may or may not exist free in nature

Examples: H, He, Na, C



Nucleus: It is central part of an atom.

Shells: Circular paths around the nucleus where electrons reside. They are also called orbits or energy levels.

No. of es in a shell = $2n^2$

Where n = no of shell.

1st shell = 2es, 2nd shell = 8es, 3rd shell = 18es.

At least 8 electron should be present in last for the stability of an atom. However if an atom has one shell like Hydrogen and Helium then 2 electrons are required for complete shell and stability.

Atomic number:

Number of protons present in the nucleus of an atom of an element.

Examples: $_{11}\text{Na}$, $_{6}\text{C}$, $_{8}\text{O}$

Atomic mass or Mass number or nucleon number: It is the sum of protons and neutrons present in the nucleus of an atom of an element.

Examples: ^{23}Na , ^{12}C , ^{16}O

O Level Chemistry Notes

Ensheng Dong



O Level Chemistry Notes:

Whispering the Secrets of Language: An Emotional Quest through **O Level Chemistry Notes**

In a digitally-driven earth wherever screens reign great and instant connection drowns out the subtleties of language, the profound techniques and psychological subtleties concealed within phrases frequently go unheard. Yet, situated within the pages of **O Level Chemistry Notes** a fascinating fictional treasure blinking with fresh thoughts, lies an extraordinary journey waiting to be undertaken. Composed by an experienced wordsmith, that wonderful opus encourages viewers on an introspective trip, delicately unraveling the veiled truths and profound impact resonating within the very cloth of each word. Within the psychological depths with this emotional review, we shall embark upon a genuine exploration of the book is primary themes, dissect their fascinating writing fashion, and fail to the powerful resonance it evokes strong within the recesses of readers hearts.

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