

Name _____

Molecular Geometries Lab Report Sheet

Formula	Lewis Structure	No of electron domains/Atp	Electron domain geometry	VSEPR Molecular geometry	Hybridization of central atom	Polar ?
CO ₂	<p>Valence 24</p> $\text{O}=\text{C}=\text{O}$ <p>2 charge clouds</p>	0	Linear 180°	Linear 	sp	NP
BF ₃	<p>Valence 32</p> $\text{F}=\text{B}-\text{F}$	0	Trigonal planar 120°		sp ²	NP
SnCl ₂	<p>Valence 24</p> $\text{Cl}=\text{Sn}-\text{Cl}$	3	Bent 120°		sp ²	P
BF ₄ ⁻	<p>Valence 40</p> $\left[\text{F}-\text{B}-\text{F} \right]^{-}$	0	Tetrahedral 109.5°		sp ³	P
AsH ₃	<p>Valence 24</p> $\text{H}-\text{As}-\text{H}$	4	Trigonal pyramidal 109.5°		sp ³	P
H ₂ S	<p>Valence 14</p> $\text{H}-\text{S}-\text{H}$	2	Bent 109.5°		sp ³	NP
NH ₂ ⁻	<p>Valence 10</p> $\left[\text{H}-\text{N}-\text{H} \right]^{-}$	2	Bent 109.5°		sp ³	P
PCl ₄	<p>Valence 40</p> $\text{F}-\text{P}-\text{Cl}$	0	Trigonal bipyre 120 + 90		sp ³ d	P

Lewis Structure And Molecular Models Lab Answers

S Ben Porath



Lewis Structure And Molecular Models Lab Answers:

Inquiries into Chemistry Michael R. Abraham, Michael J. Pavelich, 1999-05-20 The laboratory course should do more than just acquaint the students with fundamental techniques and procedures The laboratory experience should also involve the students in some of the kinds of mental activities a research scientist employs finding patterns in data developing mathematical analyses for them forming hypotheses testing hypotheses debating with colleagues and designing experiments to prove a point For this reason the student tested lab activities in *Inquiries into Chemistry 3 E* have been designed so that students can practice these mental activities while building knowledge of the specific subject area Instructors will enjoy the flexibility this text affords They can select from a comprehensive collection of structured guided inquiry experiments and a corresponding collection of open inquiry experiments depending on their perception as to what would be the most appropriate method of instruction for their students Both approaches were developed to encourage students to think logically and independently to refine their mental models and to allow students to have an experience that more closely reflects what occurs in actual scientific research Thoroughly illustrated appendices cover safety in the lab common equipment and procedures *Instructors Manual to Lab Manual* Ralph Petrucci, William Harwood, Geoffrey Herring, 2001 **Laboratory**

Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

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emphasis on handling data is supported by the Data Analysis section

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

Chemistry McGraw-Hill Staff, 2001-07 *Laboratory Manual* Jo A. Beran, 1990

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