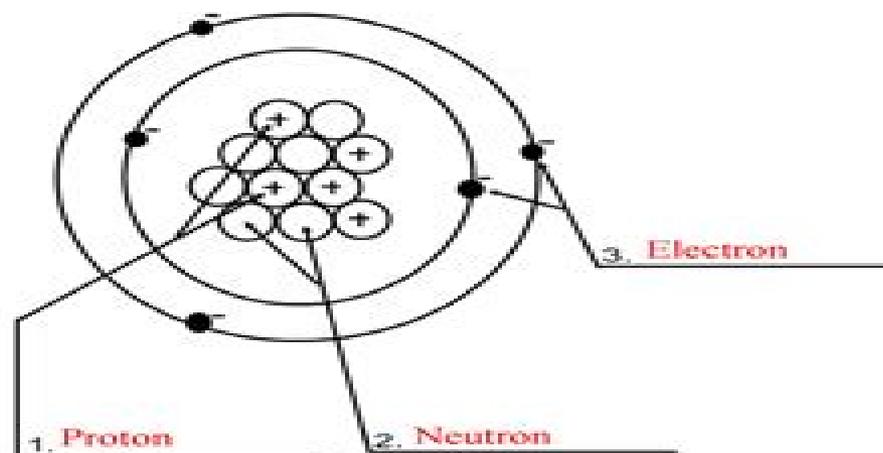


Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?
Positive (+1)

5. What type of charge does a neutron have?
Neutral (0)

6. What type of charge does an electron have?
Negative (-1)

7. Which two subatomic particles are located in the nucleus of an atom?
Protons and neutrons

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
35

9. What is the atomic number of the atom in the diagram above?
5

10. What is the atomic mass/mass number of the atom in the diagram above?
(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number

11. How many protons are in the nucleus of an atom with an atomic number of 15?
15 *number of protons is the same as the atomic number

12. How many electrons are in the nucleus of an atom with an atomic number of 20?
20

13. How many neutrons are in the nucleus of an atom with an atomic number of 25?
(use Periodic Table for mass)
30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30

14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.

15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Atomic Structure Chapter 4 Worksheet Answers

Ibrahim Sikder



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Table of Contents Atomic Structure Chapter 4 Worksheet Answers

1. Understanding the eBook Atomic Structure Chapter 4 Worksheet Answers
 - The Rise of Digital Reading Atomic Structure Chapter 4 Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Chapter 4 Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Chapter 4 Worksheet Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from Atomic Structure Chapter 4 Worksheet Answers
 - Personalized Recommendations
 - Atomic Structure Chapter 4 Worksheet Answers User Reviews and Ratings
 - Atomic Structure Chapter 4 Worksheet Answers and Bestseller Lists
5. Accessing Atomic Structure Chapter 4 Worksheet Answers Free and Paid eBooks
 - Atomic Structure Chapter 4 Worksheet Answers Public Domain eBooks
 - Atomic Structure Chapter 4 Worksheet Answers eBook Subscription Services
 - Atomic Structure Chapter 4 Worksheet Answers Budget-Friendly Options
6. Navigating Atomic Structure Chapter 4 Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Chapter 4 Worksheet Answers Compatibility with Devices
 - Atomic Structure Chapter 4 Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Chapter 4 Worksheet Answers
 - Highlighting and Note-Taking Atomic Structure Chapter 4 Worksheet Answers
 - Interactive Elements Atomic Structure Chapter 4 Worksheet Answers
8. Staying Engaged with Atomic Structure Chapter 4 Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Chapter 4 Worksheet Answers
9. Balancing eBooks and Physical Books Atomic Structure Chapter 4 Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Chapter 4 Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Chapter 4 Worksheet Answers
 - Setting Reading Goals Atomic Structure Chapter 4 Worksheet Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Atomic Structure Chapter 4 Worksheet Answers
 - Fact-Checking eBook Content of Atomic Structure Chapter 4 Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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