

Name: Date: 

## Student Exploration: Limiting Reactants

**Directions:** Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make?

2. Which ingredient limited the number of hot dogs you could make?

3. Which ingredient will you have leftovers of?

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.

To begin, make sure **H<sub>2</sub> + O<sub>2</sub> becomes H<sub>2</sub>O** is selected. The small "2" in H<sub>2</sub>, O<sub>2</sub>, and H<sub>2</sub>O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of **O<sub>2</sub> molecules** and **H<sub>2</sub> molecules** to two.

A. How many hydrogen molecules (H<sub>2</sub>) are there?

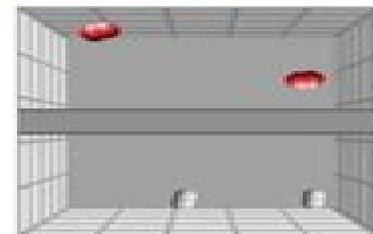
Hydrogen atoms?

B. How many oxygen molecules (O<sub>2</sub>) are there?

Oxygen atoms?

2. How many H<sub>2</sub>O molecules do you think will form when these four molecules react?

3. Click **Play** (▶). How many H<sub>2</sub>O molecules actually formed?



# Explore Learning Limiting Reactants Gizmo Answer Key

**Didier Musso**



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## **Table of Contents Explore Learning Limiting Reactants Gizmo Answer Key**

1. Understanding the eBook Explore Learning Limiting Reactants Gizmo Answer Key
  - The Rise of Digital Reading Explore Learning Limiting Reactants Gizmo Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Limiting Reactants Gizmo Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Explore Learning Limiting Reactants Gizmo Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Limiting Reactants Gizmo Answer Key
  - Personalized Recommendations
  - Explore Learning Limiting Reactants Gizmo Answer Key User Reviews and Ratings
  - Explore Learning Limiting Reactants Gizmo Answer Key and Bestseller Lists
5. Accessing Explore Learning Limiting Reactants Gizmo Answer Key Free and Paid eBooks

- Explore Learning Limiting Reactants Gizmo Answer Key Public Domain eBooks
  - Explore Learning Limiting Reactants Gizmo Answer Key eBook Subscription Services
  - Explore Learning Limiting Reactants Gizmo Answer Key Budget-Friendly Options
6. Navigating Explore Learning Limiting Reactants Gizmo Answer Key eBook Formats
    - ePub, PDF, MOBI, and More
    - Explore Learning Limiting Reactants Gizmo Answer Key Compatibility with Devices
    - Explore Learning Limiting Reactants Gizmo Answer Key Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Explore Learning Limiting Reactants Gizmo Answer Key
    - Highlighting and Note-Taking Explore Learning Limiting Reactants Gizmo Answer Key
    - Interactive Elements Explore Learning Limiting Reactants Gizmo Answer Key
  8. Staying Engaged with Explore Learning Limiting Reactants Gizmo Answer Key
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Explore Learning Limiting Reactants Gizmo Answer Key
  9. Balancing eBooks and Physical Books Explore Learning Limiting Reactants Gizmo Answer Key
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Explore Learning Limiting Reactants Gizmo Answer Key
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Explore Learning Limiting Reactants Gizmo Answer Key
    - Setting Reading Goals Explore Learning Limiting Reactants Gizmo Answer Key
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Explore Learning Limiting Reactants Gizmo Answer Key
    - Fact-Checking eBook Content of Explore Learning Limiting Reactants Gizmo Answer Key
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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