

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Matching

Match each item with the correct statement below.

Match each item with the correct statement below.

- | | |
|--------------------------------|--------------------|
| a. monatomic ion | f. cation |
| b. acid | g. binary compound |
| c. base | h. anion |
| d. law of definite proportions | i. polyatomic ion |
| e. law of multiple proportions | |

- ___ 1. consists of a single atom with a positive or negative charge
- ___ 2. atom or group of atoms having a negative charge
- ___ 3. atom or group of atoms having a positive charge
- ___ 4. tightly-bound group of atoms that behaves as a unit and carries a net charge
- ___ 5. compound composed of two different elements
- ___ 6. produces a hydrogen ion when dissolved in water
- ___ 7. produces a hydroxide ion when dissolved in water
- ___ 8. In any chemical compound, the masses of elements are always in the same proportion by mass.
- ___ 9. when two elements form more than one compound, the masses of one element that combine with the same mass of the other element are in the ratio of small, whole numbers

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ___ 10. When naming a transition metal ion that can have more than one common ionic charge, the numerical value of the charge is indicated by a ____.
- | | |
|-----------|-------------------------------------|
| a. prefix | c. Roman numeral following the name |
| b. suffix | d. superscript after the name |
- ___ 11. Which of the following correctly provides the name of the element, the symbol for the ion, and the name of the ion?
- | | |
|-----------------------------------|-------------------------------------|
| a. fluorine, F^+ , fluoride ion | c. copper, Cu^+ , cuprous ion |
| b. zinc, Zn^{2+} , zincate ion | d. sulfur, S^{2-} , sulfurous ion |
- ___ 12. The nonmetals in Groups 6A and 7A ____.
- | |
|---|
| a. lose electrons when they form ions |
| b. have a numerical charge that is found by subtracting 8 from the group number |
| c. all have ions with a -1 charge |
| d. end in <i>-ite</i> |
- ___ 13. Which of the following is NOT a cation?
- | | |
|------------------|------------------|
| a. iron(III) ion | c. Ca^{2+} |
| b. sulfate | d. mercurous ion |

Chapter 9 Practice Test Naming And Writing Chemical Formulas

Arturo Cuomo



Chapter 9 Practice Test Naming And Writing Chemical Formulas:

Unveiling the Magic of Words: A Review of "**Chapter 9 Practice Test Naming And Writing Chemical Formulas**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Chapter 9 Practice Test Naming And Writing Chemical Formulas**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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web the trail begins this section with a lengthy climb out of the columbia river gorge and eventually reaches the crest near the indian heaven wilderness a lake blessed land abounding with huckleberries next it rounds the base of mammoth mt adams elev 12 276 just north is the rugged dramatic goat rocks wilderness similar to the

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