

Soluzioni

CAPITOLO 15

VERIFICA LE TUE CONOSCENZE

VALENZA E NUMERO DI OSSIDAZIONE

- 1**
- La valenza di un elemento è il numero di elettroni che l'atomo cede, acquista o condivide quando si lega ad altri atomi.
 - La valenza massima corrisponde al numero romano che contraddistingue il gruppo, quindi, nel caso di un elemento del terzo gruppo della tavola periodica, è 3.
 - Un atomo di carbonio può formare al massimo 4 legami.
- 2** (C) di cariche reali o convenzionali associato a ciascun atomo del composto.
- 3** Regole per il calcolo del numero di ossidazione non corrette [tra parentesi, la versione corretta]:
- l'ossigeno ha sempre n.o. -2 ; [l'ossigeno nei composti ha n.o. -2 , tranne nei perossidi, nei superossidi e nel difluoruro di ossigeno, dove ha rispettivamente n.o. -1 , $-1/2$, $+2$];
 - l'idrogeno ha n.o. $+1$ nei composti e $+2$ nella molecola H_2 ; [l'idrogeno nei composti ha n.o. $+1$ quando è legato a un non metallo e -1 quando è legato a un metallo].

SCRIVERE LE FORMULE DEI COMPOSTI A PARTIRE DAI NUMERI DI OSSIDAZIONE

- 4** Na_2O : in questo modo la somma algebrica dei n.o. è uguale a zero: $2 \cdot (+1) + (-2) = 0$.
- 5** $Ca_3(PO_4)_2$
- 6**
- SO_4^{2-} ; carica dello ione = -2
 - ClO_7^- ; carica dello ione = -1
- 7** La formula Cl_2Ba è sbagliata, in quanto si scrive a sinistra l'atomo con n.o. positivo (Ba_2Cl_2).
- 8** È corretta soltanto: d. SO_2 [tra parentesi, le formule corrette]:
- HS (H_2S)
 - Fe_3S_2 (Fe_2S_3)
 - LiS_2 (Li_2S)

LA NOMENCLATURA DEI COMPOSTI BINARI

- 9** Un composto binario è formato da due elementi diversi. I composti binari si dividono in:
- composti contenenti ossigeno: ossidi (es. ZnO), perossidi (es. H_2O_2), superossidi;
 - composti contenenti idrogeno: idruri (es. NaH), idracidi (es. HF);
 - sali binari (es. $NaCl$, $CaCl_2$).
- 10** In molti composti binari si ha la desinenza *-ure*; tale desinenza non compare negli ossidi.
- 11** La notazione di Stock si utilizza quando gli elementi che formano un composto hanno più di un numero di ossidazione. Comporta l'eliminazione dei prefissi numerici e l'introduzione del numero di ossidazione dell'elemento, come numero romano tra parentesi posto dopo il nome dell'elemento. Si ritiene che in questo modo il nome che contraddistingue l'elemento risulti più breve e più chiaro.
- | | | |
|-------------------|----------|-----------------------|
| (n.o. Fe = $+2$) | $FeBr_2$ | bromuro di ferro(II) |
| (n.o. Fe = $+3$) | $FeBr_3$ | bromuro di ferro(III) |
- 12** Lo ione ossido (O^{2-}) è formato da un atomo di O con n.o. = -2 .
Lo ione perossido (O_2^{2-}) è formato da due atomi di O entrambi con n.o. = -1 , legati da legame covalente.
Lo ione superossido (O_2^-) è formato da due atomi di O con n.o. medio = $-1/2$, legati da legame covalente.
Un ossido acido è formato da un non metallo e ossigeno, un ossido basico è formato da un metallo e ossigeno.
- 13**
- Idrogeno e metallo: idruri (NaH , LiH);
 - idrogeno e non metallo dei gruppi 16 e 17: idracidi (HCl , HF , HBr);
 - metallo e non metallo: sali binari ($NaCl$, $FeCl_2$).
- 14** (B) cianidrico.
- 15** K_2O : ossido di potassio;
 SO_2 : anidride solforosa (n.o. S = $+4$).
 K_2O si chiama *ossido* perché è un ossido basico (metallo e ossigeno).
 SO_2 si chiama *anidride* perché è un ossido acido (non metallo e ossigeno).

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640 domande e 840 risposte sulle reti LAN e WAN Marco Paganini, 2009-09-01 Un intero percorso di preparazione alla Certificazione Cisco CCNA 640 802 versione 4 preparato da un docente esperto Praticamente bilingue con tutti i termini chiave sempre riportati in italiano e inglese e con gli acronimi solitamente affiancati dalla versione espansa per maggior chiarezza Ricco di esempi spiegazioni riferimenti approfondimenti tabelle riassuntive Segnala anche le piccole sviste o i punti critici presenti secondo l'autore nel materiale originale in inglese Strutturato a domande e risposte per facilitare l'apprendimento dividendo la materia in pillole formative sui singoli argomenti Approccio totalmente top down le reti in generale i protocolli dal livello 7 al livello 2 le interfacce e i cablaggi livello 1 gli apparati Uno zoom sui Router sugli Switch sugli Access Point sulla sicurezza ICT e sulle ACL Access Control List Comprende due accurate Command Reference dei comandi dei Router e degli Switch a livello CCNA con alcune utili integrazioni con i prompt e tutti i parametri commentati ed esemplificati il libro che pu aiutare docenti e studenti che seguono i percorsi delle Cisco Academies per incrociare approfondire e integrare i materiali didattici online e chi pi esperto si affida alla modalit self study **Bible as Notepad** Liv Ingeborg Lied, Marilena Maniaci, 2018-09-10 The present volume provides a comparative look at the contents and layout features of secondary annotations in biblical manuscripts across linguistic traditions Due to the privileged focus on the text in the columns these annotations and the practices that produced them have not received the scholarly attention they deserve The vast richness of extant verbal and figurative notes accompanying the biblical texts in the intercolumns and margins of the manuscript pages have thus been largely overlooked The case studies gathered in this volume explore Jewish and Christian biblical manuscripts through the lens of their annotations addressing the various relationships between the primary layer of text and the secondary notes and exploring the roles and functions of annotated manuscripts as cultural artifacts By approaching biblical manuscripts as potential notepads the volume offers theoretical reflection and empirical analyses of the ways in which secondary notes may shed new light on the development and transmission of text traditions the shifting engagement with biblical manuscripts over time as well as the change of use and interpretation that may result from the addition of the notes themselves **Hacker all'attacco. La tua rete è a rischio** R. Russell, 2003 **Le nuove professioni del web** Giulio Xhaet, 2012-09-06T10:38:14+02:00 Giulio Xhaet ci accompagna quasi per mano attraverso un inventario completo di tutte le nuove professionalit che Internet ha creato nel mondo Ma l'aspetto realmente inedito di quest'opera la prospettiva con cui ogni mestiere viene raccontato ai lettori Non sono solo le competenze a essere descritte ma le attitudini le sensibilit i talenti le predisposizioni obbligatorie per ogni professione analizzata Perch qui si gioca la vera cifra di chi lavora con Internet ed quello che molto spesso pu fare la differenza tra professionalit e improvvisazione Lasciate che le prossime pagine vi orientino nel vastissimo panorama che le nuove professioni del Web ci offrono oggi e domani C di che stare tranquilli a leggere quali e quante nuove professioni stanno aspettando di incontrarvi Ma nessun incontro sar realmente

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