

Practice Test #21: Nuclear Chemistry

1. Of the following processes, which one does not change the atomic number?

- A) alpha emission ${}^z_A X \rightarrow {}^{z-4}_{A-4} Y + {}^4_2\alpha$
- B) beta emission ${}^z_A X \rightarrow {}^{z+1}_A Y + {}^0_{-1}\beta$
- C) electron capture ${}^z_A X + {}^0_{-1}e(\text{orbital electron}) \rightarrow {}^{z-1}_A Y$
- D) gamma emission ${}^z_A X^* \rightarrow {}^z_A X + {}^0_0\gamma$ (only this nuclear reaction does not change the atomic number)
- E) positron emission ${}^z_A X \rightarrow {}^{z-1}_A Y + {}^0_{+1}e$

2. In balancing the nuclear reaction below, the identity of element "E" is determined from its...



- A) atomic weight.
- B) mass number.
- C) atomic number.
- D) number of neutrons.
- E) number of electrons.

3. The largest number of stable nuclei have an _____ number of protons and an _____ number of neutrons.

- A) even, even
- B) odd, odd
- C) even, odd
- D) odd, even

4. The reaction below is an example of _____ decay.



- A) alpha
- B) beta
- C) gamma
- D) positron

Study Nuclear Chemistry Answer Key

Mike Jess



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General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

Chemistry, Student Study Guide John A. Olmsted, Gregory M. Williams, 2002-01-04 This third edition continues to innovate by providing students with an integrated and modern approach to the subject The text emphasizes the modern tools of chemistry while incorporating historical evidence and its unique molecular quantitative emphasis is further reinforced by an integrated media package developed by the authors Also of benefit is the just in time presentation of key content only providing details once they are needed While key topics and analytical techniques have been updated there is now an additional third chapter on chemical equilibrium The authors have also developed an expanded and more integrated problem solving emphasis that now incorporates a 4 step strategy throughout complete with text icons The whole is backed by a range of supplements including a new illustration program a tutorial CD interactive learningware an extensive Web CT component an instructor s resource CD and a solution CD

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