

Name : \_\_\_\_\_

Date : \_\_\_\_\_

# Acids and Bases Problems

- Without doing any calculations, identify each of the following solutions as acidic or basic:
  - $[\text{H}_3\text{O}^+] = 3.4 \times 10^{-8} \text{ M}$  \_\_\_\_\_
  - $[\text{OH}^-] = 1.8 \times 10^{-5} \text{ M}$  \_\_\_\_\_
  - $[\text{OH}^-] = 6.5 \times 10^{-12} \text{ M}$  \_\_\_\_\_
  - $[\text{H}_3\text{O}^+] = 2.6 \times 10^{-4} \text{ M}$  \_\_\_\_\_
- A solution of HBr has  $[\text{H}_3\text{O}^+] = 4.5 \times 10^{-3} \text{ M}$ . Calculate the pH of this solution.

- Complete the following table.

pH	$[\text{H}_3\text{O}^+]$	$[\text{OH}^-]$	Acidic, basic, or neutral?
	$5.4 \times 10^{-4}$		
	$1.3 \times 10^{-5}$		
10.75	$1.8 \times 10^{-11}$	$5.6 \times 10^{-4}$	

- If the pH of a solution is 10.3, what is the  $[\text{H}^+]$  concentration?
- If the  $[\text{H}^+]$  is  $2.1 \times 10^{-12} \text{ M}$   $\text{HClO}_4$ , what is the pH? Is the solution ACIDIC, BASIC, or NEUTRAL?
- Calculate the pH if the  $[\text{OH}^-]$  concentration is  $5.9 \times 10^{-1} \text{ M}$ ? Is the solution ACIDIC, BASIC, or NEUTRAL?

# Acids Bases Chemistry Test Answers

**M Mosston**



**Acids Bases Chemistry Test Answers:**

## Unveiling the Magic of Words: A Review of "**Acids Bases Chemistry Test Answers**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Acids Bases Chemistry Test Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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