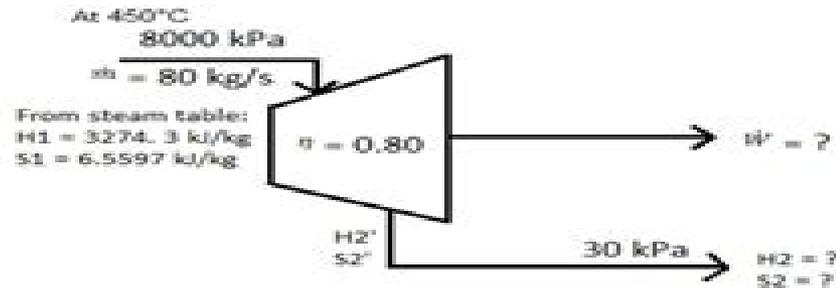


Introduction to Chemical Engineering Thermodynamics
Problem Set

1. A turbine operates adiabatically with superheated steam entering at 450°C and 8,000 kPa with a mass flowrate of 80 kg/s. The exhaust pressure is 30 kPa and the turbine efficiency is 0.80. Determine the power output of the turbine and the enthalpy and entropy of the exhaust steam.

Given:



Required:

- a. \dot{W} , power output of the turbine
- b. H_2 , enthalpy of exhaust stream
- c. S_2 , entropy of exhaust stream

Solution:

From steam table under superheated steam, Table F.2, for inlet conditions:

$$H_1 = 3274.3 \frac{\text{kJ}}{\text{kg}}$$

$$S_1 = 6.5597 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

For discharge conditions:

$$H_2^L = 289.302 \frac{\text{kJ}}{\text{kg}}, H_2^V = 2625.4 \frac{\text{kJ}}{\text{kg}}$$

$$S_2^L = 0.9441 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}, S_2^V = 7.7695 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$$

If the expansion at 30 kPa is isentropic, then: $S_2^L = S_1 = 6.5597 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}$. Steam with this entropy at 10 kPa is wet, yielding the equation:

$$S_2^L = S_2^L + x_2'(S_2^V - S_2^L)$$

Substituting the values:

$$6.5597 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} = 0.9441 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} + x_2' \left(7.7695 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} - 0.9441 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \right)$$

$$x_2' = 0.8228$$

This is the quality (fraction vapor) of the discharge stream at point 2'. The enthalpy H_2^L is also given by the equation:

$$H_2^L = H_2^L + x_2'(H_2^V - H_2^L)$$

Thus,

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M Mark



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Table of Contents Solved Problems In Chemical Engineering Thermodynamics

1. Understanding the eBook Solved Problems In Chemical Engineering Thermodynamics
 - The Rise of Digital Reading Solved Problems In Chemical Engineering Thermodynamics
 - Advantages of eBooks Over Traditional Books
2. Identifying Solved Problems In Chemical Engineering Thermodynamics
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solved Problems In Chemical Engineering Thermodynamics
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solved Problems In Chemical Engineering Thermodynamics
 - Personalized Recommendations

- Solved Problems In Chemical Engineering Thermodynamics User Reviews and Ratings
- Solved Problems In Chemical Engineering Thermodynamics and Bestseller Lists
- 5. Accessing Solved Problems In Chemical Engineering Thermodynamics Free and Paid eBooks
 - Solved Problems In Chemical Engineering Thermodynamics Public Domain eBooks
 - Solved Problems In Chemical Engineering Thermodynamics eBook Subscription Services
 - Solved Problems In Chemical Engineering Thermodynamics Budget-Friendly Options
- 6. Navigating Solved Problems In Chemical Engineering Thermodynamics eBook Formats
 - ePub, PDF, MOBI, and More
 - Solved Problems In Chemical Engineering Thermodynamics Compatibility with Devices
 - Solved Problems In Chemical Engineering Thermodynamics Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Solved Problems In Chemical Engineering Thermodynamics
 - Highlighting and Note-Taking Solved Problems In Chemical Engineering Thermodynamics
 - Interactive Elements Solved Problems In Chemical Engineering Thermodynamics
- 8. Staying Engaged with Solved Problems In Chemical Engineering Thermodynamics
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solved Problems In Chemical Engineering Thermodynamics
- 9. Balancing eBooks and Physical Books Solved Problems In Chemical Engineering Thermodynamics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solved Problems In Chemical Engineering Thermodynamics
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Solved Problems In Chemical Engineering Thermodynamics
 - Setting Reading Goals Solved Problems In Chemical Engineering Thermodynamics
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solved Problems In Chemical Engineering Thermodynamics
 - Fact-Checking eBook Content of Solved Problems In Chemical Engineering Thermodynamics

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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