

## THEIR EYES WERE WATCHING GOD Standards Focus Activities

### Chapters Nineteen and Twenty

### Standards Focus: Theme

**Theme** is the central idea in a work of literature. The theme of a piece of literature should not be confused with the subject of the work, but rather, it is a general statement about life or human nature. Most themes are not completely obvious and must be inferred by the reader. The reader must take a good look at the characters, plot, setting, mood, even the title, and how they work together, to understand and recognize the reasons Hurston wrote this particular piece of literature.

Directions: For numbers 1-5, a theme from the novel has been chosen. Find a quote from the text which best illustrates each theme. For numbers 6-10, a quote has been taken directly from the text. Write what additional themes these quotes reveal or suggest. **Remember that theme is not a one-word response, but rather, a complete statement of opinion.** The chapter number has been given in order for you to look up the reference if necessary.

1. The quest for identity and autonomy is not easy.

Example:

2. Although family and friends may have the best intentions, they should not decide your life's path.

Example:

3. Folk culture and religion is an important aspect of the African-American community.

Example:

4. Everyone, regardless of age, race, or gender has a right to achieve happiness.

Example:

5. Women often have a more difficult time in life because of gender roles and expectations.

Example:

# Standards Focus Dialogue Chapter One Answers

**Guillaume Favre**



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{CO}_2 \rightarrow \text{Fe}_2\text{O}_3 + \text{C}$   
oxidation  $\text{CO} + \text{CO}_2 + 2\text{e}^-$  reduction  $\text{Fe} + 2\text{e}^- \rightarrow \text{Fe}^{2+}$   $24 \text{ 3 Ag} + \text{Ni}^{3+} \rightarrow 3 \text{ Ag} + \text{Ni}$  oxidation reduction  $25 \text{ Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$  oxidation reduction  $26 \text{ O}_2$

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 $\text{a} \text{ I}_2 + \text{O}_5 \rightarrow \text{I}_2\text{O}_5$   
 $\text{b} \text{ 5CO} + \text{I}_2 \rightarrow \text{5CO}_2 + \text{I}_2$   
 $\text{c} \text{ O}_5 + \text{C} \rightarrow \text{CO}_2$

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