

# COMPUTATIONAL SCIENCE & ENGINEERING

## Building Blocks



# Computational Science And Engineering Springer

**LL Leslie**



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web write down the equilibrium equation you are investigating using the information provided by the introduction within the virtual lab activity equilibrium reaction  $co_2 + h_2o \rightleftharpoons h_2co_3$  select each control test tube to compare the concentrations of

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web the study of reaction rates will lead you into the study of chemical equilibrium in this semester you will also study electrochemistry from batteries to electroplating and you will learn about acids bases and salts unit 15 is a brief study of the atomic nucleus which your teacher may introduce at any time during the year

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web equilibrium moles  $hscn = 2.00 \times 10^{-5} \text{ mol}$   $3.00 \times 10^{-6} \text{ mol}$   $1.70 \times 10^{-5} \text{ mol}$   $hscn$  knowing the number of moles of  $fe^{3+}$  and  $hscn$  present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species  $mol\ fe^{3+} = 1.70 \times 10^{-5} \text{ mol}$   $mol\ fe^{3+} = 8.50 \times 10^{-4} \text{ m}$

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