

# STUDY GUIDE

## SECTION 1.1 - 1.4

- **matter**: everything that takes up space + has a mass

### Properties of matter



### ① Physical changes:

- Phase
- Size
- Texture

→ Reversible

### ② Chemical changes:

- Color change
- New substance is formed

→ Irreversible

### Pure substance vs. Mixture

#### MOLECULE (pure substances)

##### Compounds

2 or more ≠ atoms  
(eg:  $H_2O$ ;  $Na_2O$ )

##### Elements

1 atom  
(eg:  $H_2$ ,  $O_2$ )

- **Pure substance**: constant composition + definite properties

# Chemistry Midterm Study Guide

**Cathy Duffy**



## **Chemistry Midterm Study Guide:**

Student Study Guide Peter L. Kresan, Reed Mencke, Frank Press, Raymond Siever, 2003-09-25 This reconceptualization of the text *Understanding Earth* reflects the fundamental changes in the field of physical geology over the past several years

*PSI Study Guide for Organic Chemistry* by Douglas C. Neckers, Michael P. Doyle E. C. Blossey, 1977 McGraw-Hill's *500 College Chemistry Questions* David E. Goldberg, 2012-09-07 500 Ways to Achieve Your Best Grades We want you to succeed on your college chemistry midterm and final exams That's why we've selected these 500 questions to help you study more effectively use your preparation time wisely and get your best grades These questions are similar to the ones you'll find on a typical college exam so you will know what to expect on test day Each question includes comprehensive explanations in the answer key Whether you have been studying all year or are doing a last minute review McGraw Hill's 500 College Chemistry Questions will help you achieve the final grade you desire Sharpen your subject knowledge and build your test taking confidence with 500 essential college chemistry questions with answers Clear solutions in the answer key for every problem Coverage from atomic mass to electrochemistry

*Teaching and Learning STEM* Richard M. Felder, Rebecca Brent, 2016-02-22 Rethink traditional teaching methods to improve student learning and retention in STEM Educational research has repeatedly shown that compared to traditional teacher centered instruction certain learner centered methods lead to improved learning outcomes greater development of critical high level skills and increased retention in science technology engineering and mathematics STEM disciplines *Teaching and Learning STEM* presents a trove of practical research based strategies for designing and teaching STEM courses at the university community college and high school levels The book draws on the authors extensive backgrounds and decades of experience in STEM education and faculty development Its engaging and well illustrated descriptions will equip you to implement the strategies in your courses and to deal effectively with problems including student resistance that might occur in the implementation The book will help you Plan and conduct class sessions in which students are actively engaged no matter how large the class is Make good use of technology in face to face online and hybrid courses and flipped classrooms Assess how well students are acquiring the knowledge skills and conceptual understanding the course is designed to teach Help students develop expert problem solving skills and skills in communication creative thinking critical thinking high performance teamwork and self directed learning Meet the learning needs of STEM students with a broad diversity of attributes and backgrounds The strategies presented in *Teaching and Learning STEM* don't require revolutionary time intensive changes in your teaching but rather a gradual integration of traditional and new methods The result will be continual improvement in your teaching and your students learning More information about *Teaching and Learning STEM* can be found at <http://educationdesignsinc.com> book including its preface foreword table of contents first chapter a reading guide and reviews in 10 prominent STEM education journals

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