

Covalent Bonding Worksheet

Covalent bonding occurs when two or more **NON_METALS** share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H ₂ hydrogen is diatomic: $\text{H} + \text{H} \longrightarrow \text{H} \cdot \cdot \text{H} \longrightarrow \text{H}-\text{H} \quad \text{Single Bond}$
2.) F ₂ fluorine is diatomic: $\text{F} + \text{F} \longrightarrow \cdot \cdot \text{F} \cdot \cdot \cdot \cdot \text{F} \cdot \cdot \longrightarrow$
3.) O ₂ oxygen is diatomic: $\text{O} + \text{O} \longrightarrow \quad \quad \quad \longrightarrow \text{O}=\text{O} \quad \text{Double Bond}$
4.) N ₂ nitrogen is diatomic. Is this a triple bond? $\text{N} + \text{N}$
5.) BF ₃ you need 3 fluorine atoms here: $\text{B} + \text{F}$
6.) Ammonia NH ₃ hint: how many hydrogen atoms are needed? $\text{N} + \text{H}$
7.) Carbon dioxide CO ₂ $\text{O} + \text{C} + \text{O}$
8.) Methane CH ₄ careful here 4 hydrogen atoms needed $\text{C} + \text{H}$
9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!! $\text{H} + \text{O} + \text{H}$
10.) SO ₂ hint: one pair of electrons from sulfur must be slit up for this one to work. $\text{O} + \text{S} + \text{O}$

Chapter 8 Covalent Bonding Worksheet Answers

Keith Taber



Chapter 8 Covalent Bonding Worksheet Answers:

Chemical Misconceptions Keith Taber, 2002 Part 2 provides strategies for dealing with some of the misconceptions that students have by including ready to use classroom resources [Educart ICSE Class 10 One-shot Question Bank 2026 Chemistry \(strictly for 2025-26 boards\)](#) Sir Tarun Rupani, 2025-07-12 Fast track your Chemistry revision with this exam ready resource This One shot Question Bank by Sir Tarun Rupani is designed to help ICSE Class 10 students revise the complete Chemistry syllabus quickly and thoroughly It simplifies theory boosts numerical accuracy and ensures strong exam practice all aligned with the 2025 26 ICSE syllabus Key Features Strictly Based on ICSE 2025 26 Curriculum Complete chapter coverage including Periodic Table Chemical Bonding Acid Base Organic Chemistry and more One shot Format Each chapter includes concise concept notes chemical equations reactions and key diagrams for quick recall Complete Coverage of Question Types Includes objective short long answers equation based numerical and reasoning questions Chapterwise PYQs Included Practice with previous years ICSE board questions to understand trends and improve retention Solved Answers in ICSE Format Clear well structured solutions using proper units chemical symbols and balanced equations Smart Revision Focus Special tips to avoid common mistakes in writing reactions balancing equations and attempting numericals Why Choose This Book This Chemistry One shot by Sir Tarun Rupani is built for smart preparation whether you re revising at the last minute or practising throughout the term It helps you approach each question with clarity confidence and the precision needed to score high in the 2026 ICSE board exam **Addison-Wesley Science Insights** ,1996 [Books in Print Supplement](#) ,2002 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction [MnM_POW-Science-PM-10 \(Updated\)](#) Vibha Arora, Anju Sachdeva, Sushma Sardana, MnM_POW Science PM 10 Updated [Exploring Earth and Space](#) Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter *Chapter Resource 2 Chemistry of Life Biology* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 **Earth Science** Samuel N. Namowitz, Nancy E. Spaulding, 1989 **Discover Science: Teacher's annotated edition** ,1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today s technological society [Holt Science and Technology](#) Holt, Rinehart and Winston Staff, 2001

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