

Introduction to Solutions

Instructions: Circle the substances below that would be classified as **solutions**.

<p>Apple Juice</p> 	<p>Salad</p> 	<p>Dirt</p> 	<p>Lemonade</p> 
<p>Coffee with Sugar</p> 	<p>Pure Water (H₂O)</p> 	<p>Cookie</p> 	<p>Soda Pop</p> 

Instructions: Use the choices in the word bank below to answer the following questions:

Aqueous Electrolyte Solution Water Solute Solvent Solvation

- 1) **Solute** The substance that is dissolved in a solution (i.e. sugar in water)
- 2) **Solution** A homogenous mixture of two or more substances.
- 3) **Electrolyte** Solution that can conduct an electric current due to dissolved ions.
- 4) **Solvation** The process of dissolving. Solute particles are surrounded by solvent.
- 5) **Aqueous** A solution in which the solvent is water (H₂O)
- 6) **Solvent** The substance that is doing the dissolving in a solution (i.e. water)
- 7) **Water** This substance is often referred to as the "universal solvent."

Solutions Worksheets For Chemistry

Steve DeRoy



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