

NB: Iron cross membrane as:

- Ferrous state Fe^{+2} .
- Ferric state Fe^{+3} conjugated with protein.

▣ occurs at → duodenum & proximal part of jejunum.

▣ absorption is enhanced → When iron stores in body are depleted.

NB: When adequate quantity of iron is stored, absorption is ↓.

▣ **Only ferrous** (and not ferric) form of iron **is absorbed**.

▣ Ferric iron is reduced to ferrous iron by ferric reductase (an enzyme present on surface of enterocytes).

▣ Ferrous iron in intestinal lumen binds to mucosal cell protein, called **divalent metal transporter-1 (DMT-1)**.

▣ This bound iron is then transported into mucosal cell.

▣ rest of the unabsorbed iron is excreted.

▣ **Inside intestinal cell:**

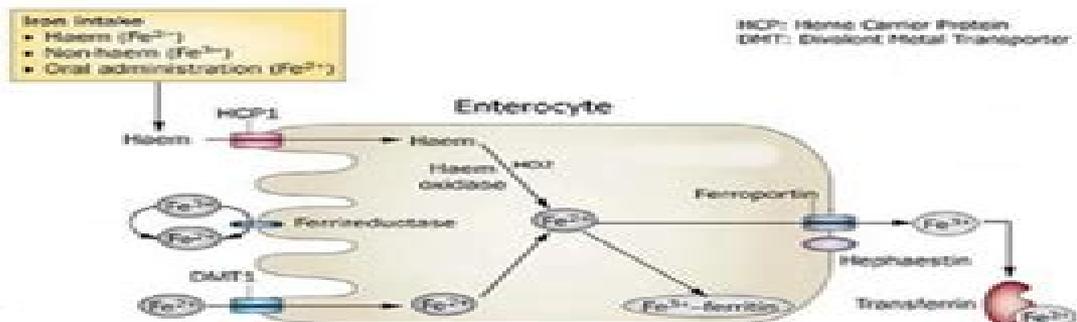
- iron is oxidized to ferric state.
- and **combined with apoferritin** to form **ferritin** (main storage form in intestinal cells). kept temporarily in the mucosal cell.

NB: in case of anemia, iron is further absorbed into bloodstream.

▣ If transferrin is saturated with iron, any iron accumulated in mucosal cell is lost **when cell is desquamated**.

▣ fraction of iron absorbed & retained is decided by iron status.

▣ When iron is in excess, absorption is reduced; this is basis of "mucosal block".



Biochemistry Easy

David Baud



Biochemistry Easy:

Biochemistry Made Easy Jon Adams, Biochemistry Made Easy unravels the complex and captivating world of biochemistry transforming it into a story rich with characters and narratives that make abstract concepts graspable Crafted for a diverse audience including students educators and professionals this book demystifies the dense terminology and intricate processes of biochemistry through deep analogies and vivid examples making the science accessible and engaging

Chapter Contents Chapter 1 The Alphabet of Life Amino Acids Delve into the basic building blocks of proteins and how they shape the functions of living organisms Chapter 2 The Cellular Factory Enzyme Kinetics Explore the mechanisms that drive the rates of biochemical reactions crucial for understanding metabolic processes Chapter 3 The Energy Currency ATP and Bioenergetics Unpack the process of energy storage and expenditure that powers cellular functions Chapter 4 Lipids and Membranes The Body's Insulators Learn about the role of lipids in forming cellular barriers and facilitating energy storage Chapter 5 The Genetic Library DNA and Inheritance Investigate the molecular basis of genetics and how traits are passed from one generation to the next Chapter 6 The Central Dogma Transcription and Translation Understand the flow of genetic information from DNA to RNA to protein a cornerstone of molecular biology Chapter 7 Metabolic Pathways The Body's Highways Discover the routes taken by molecules in a cell to sustain life featuring key pathways like glycolysis and the Krebs cycle Chapter 8 Signaling Molecules The Communication Network Examine how cells communicate with each other to coordinate actions across an organism

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Clinical Biochemistry Made Easy Dr Uthayasooriya Amarasena,2025-08-21 What is Biochemistry Core Definition Biochemistry is the branch of science that explores the chemical processes and substances that occur within living organisms It bridges biology and chemistry revealing how molecular interactions sustain life from DNA replication to enzyme action from nutrient metabolism to cellular signaling Biochemistry is the language of life spoken in molecules and understood through reactions Scope and Significance Biochemistry underpins nearly every aspect of medicine and physiology Cellular Function Explains how cells generate energy replicate and respond to stimuli Disease Mechanisms Reveals molecular defects in conditions like diabetes cancer and genetic disorders Diagnostics Forms the basis of lab tests e g liver enzymes blood glucose ABG analysis Therapeutics Guides drug design enzyme replacement and nutritional interventions Preface Clinical Biochemistry Made Easy By Dr Uthayasooriya Amarasena Clinical biochemistry is often perceived as a labyrinth of reactions pathways and molecular jargon daunting to many medical undergraduates and nursing students alike This series was born from a simple yet profound question How can we make biochemistry not only understandable but memorable clinically relevant and emotionally resonant for every learner who opens the book As an

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The Chemistry of Life John Stanley Durrant Bacon,1944

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Biochemistry and Molecular Biology William H. Elliott,Daphne C. Elliott,2005 A concise introductory textbook in biochemistry and molecular biology for life sciences students taking a first course in the topic Professor William Elliott from University of Adelaide Dr Daphne Elliott formerly at Flinders University

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