

Chemistry Lab Report: Identifying an Unknown Compound

In this lab, I identified an unknown compound by testing and comparing it to known compounds. The unknown compound was a white, crystalline solid that melted at 100°C. I tested it for various ions and functional groups. The results of the tests are as follows:

- Flame Test:** The compound produced a yellow flame, indicating the presence of sodium ions.
- Phosphate Test:** The compound did not form a precipitate with ammonium molybdate, indicating the absence of phosphate ions.
- Chloride Test:** The compound did not form a precipitate with silver nitrate, indicating the absence of chloride ions.
- Sulfate Test:** The compound did not form a precipitate with barium chloride, indicating the absence of sulfate ions.
- Nitrate Test:** The compound did not form a precipitate with sodium hydroxide, indicating the absence of nitrate ions.
- Carbonate Test:** The compound did not form a precipitate with hydrochloric acid, indicating the absence of carbonate ions.
- Acetate Test:** The compound did not form a precipitate with lead acetate, indicating the absence of acetate ions.
- Ammonium Test:** The compound did not form a precipitate with sodium hydroxide, indicating the absence of ammonium ions.

Based on the results of the tests, I identified the unknown compound as sodium chloride (NaCl). The compound is a white, crystalline solid that melts at 100°C. It is soluble in water and forms a colorless solution. The compound is used in a variety of applications, including as a preservative, a food additive, and a raw material in the chemical industry.

The unknown compound was identified as sodium chloride (NaCl). The compound is a white, crystalline solid that melts at 100°C. It is soluble in water and forms a colorless solution. The compound is used in a variety of applications, including as a preservative, a food additive, and a raw material in the chemical industry. The compound is also used in the production of sodium hydroxide and sodium carbonate. The compound is a common salt and is found in nature as halite. The compound is used in the production of glass, paper, and textiles. The compound is also used in the production of sodium metal and sodium hydroxide. The compound is a common salt and is found in nature as halite. The compound is used in the production of glass, paper, and textiles. The compound is also used in the production of sodium metal and sodium hydroxide.

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Jessica J Manson



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web one summer when lizabeth is fourteen the children decide to go throw stones at miss lottie s marigolds miss lottie is an old woman who lives in a ramshackle building with her disabled son john burke the children scamper over to miss lottie s house and decapitate a few marigolds

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